Typical diseases

The search for pharmaceutical drugs used to be rather straight forward until recent times:

A wealth of information about the disease, its causes, and the clinical symptoms were readily available. Thus the starting point for the pharmacological therapy was known.

Example: inhibition of an enzyme

Thus the target was fixed. Frequently, experience with existing medications was available. Therefore a *valid target* or at least a *drugable target* was present.

→ The *target* undergoes a change of its activity caused by the drug

Often, multiple targets can be exploited to treat the same disease. → Therapeutic Categories

Therapeutic Categories

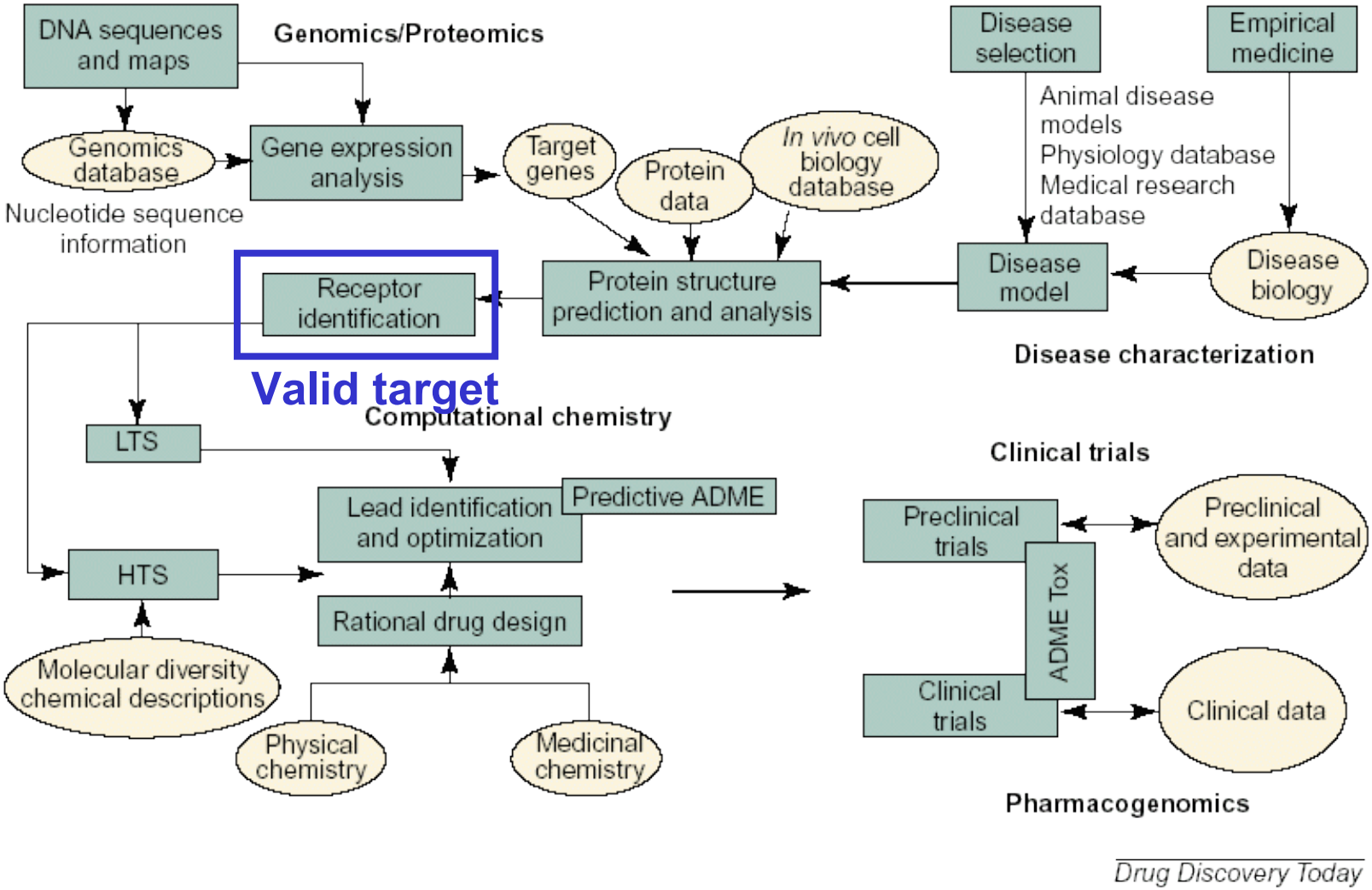
Grouping drugs under the aspect of their pharmacological and therapeutic application results in about 200 categories:

ACE Inhibitor
Adrenocortical Suppressant
Adrenocorticotropic Hormones
Aldose Reductase Inhibitors
Aldosterone Antagonists
α-adrenergic Agonists
α-adrenergic Blockers
α-Glucosidase Inhibitors
Anabolic Streroids
Analgesic, Dental
Analgesic, Narcotic
Analgesic, Non-narcotic
Androgens
Anesthetics, Inhaled
Anesthetics, Intravenous
Anesthetics, Local
Angiotensin II Antagonists
Anorexics

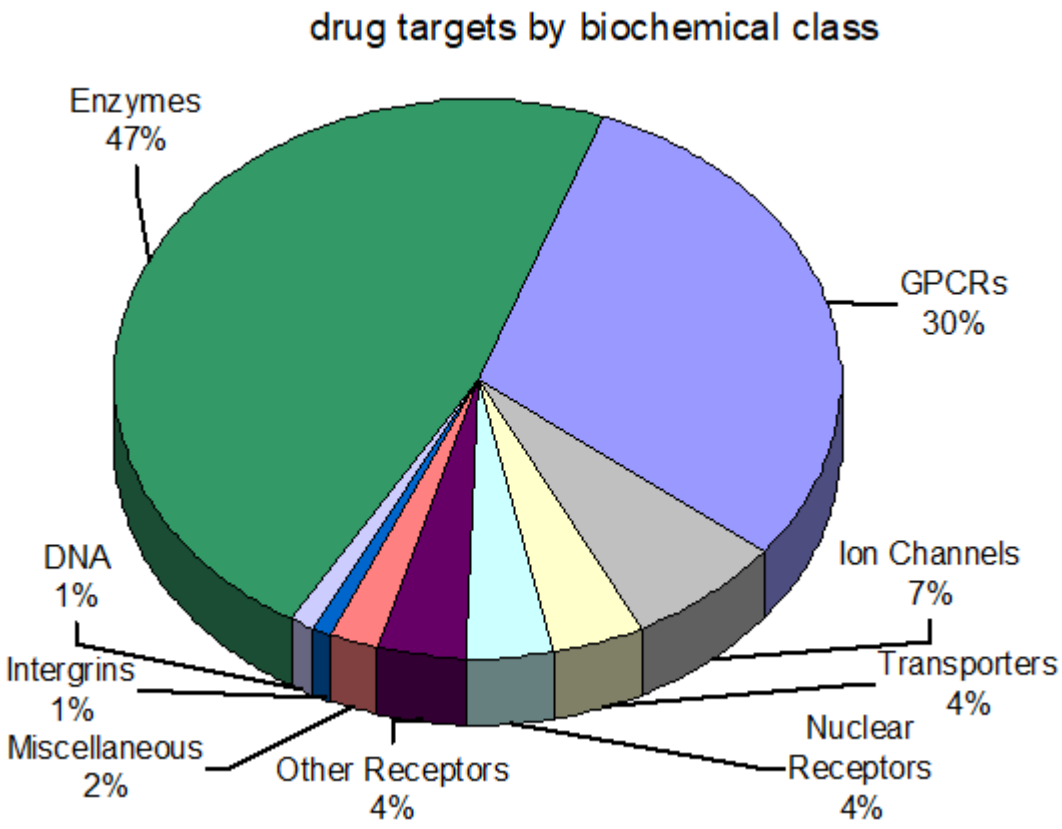
see the corresponding section in the Merck Index

In most cases it is not obvious to recognize the treated disease from a therapeutic class (at least for non-medical persons), sometimes not even the actual molecular target(s).

Flow of information in a drug discovery pipeline



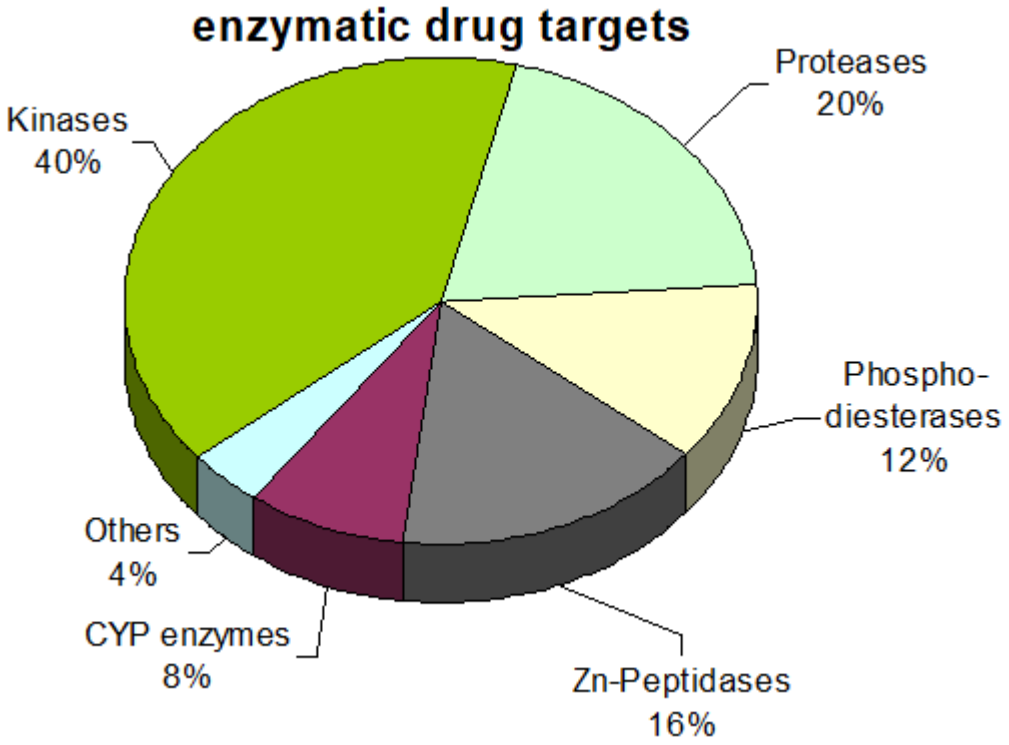
typical targets



Fractional content of marketed drugs according to their biochemical targets

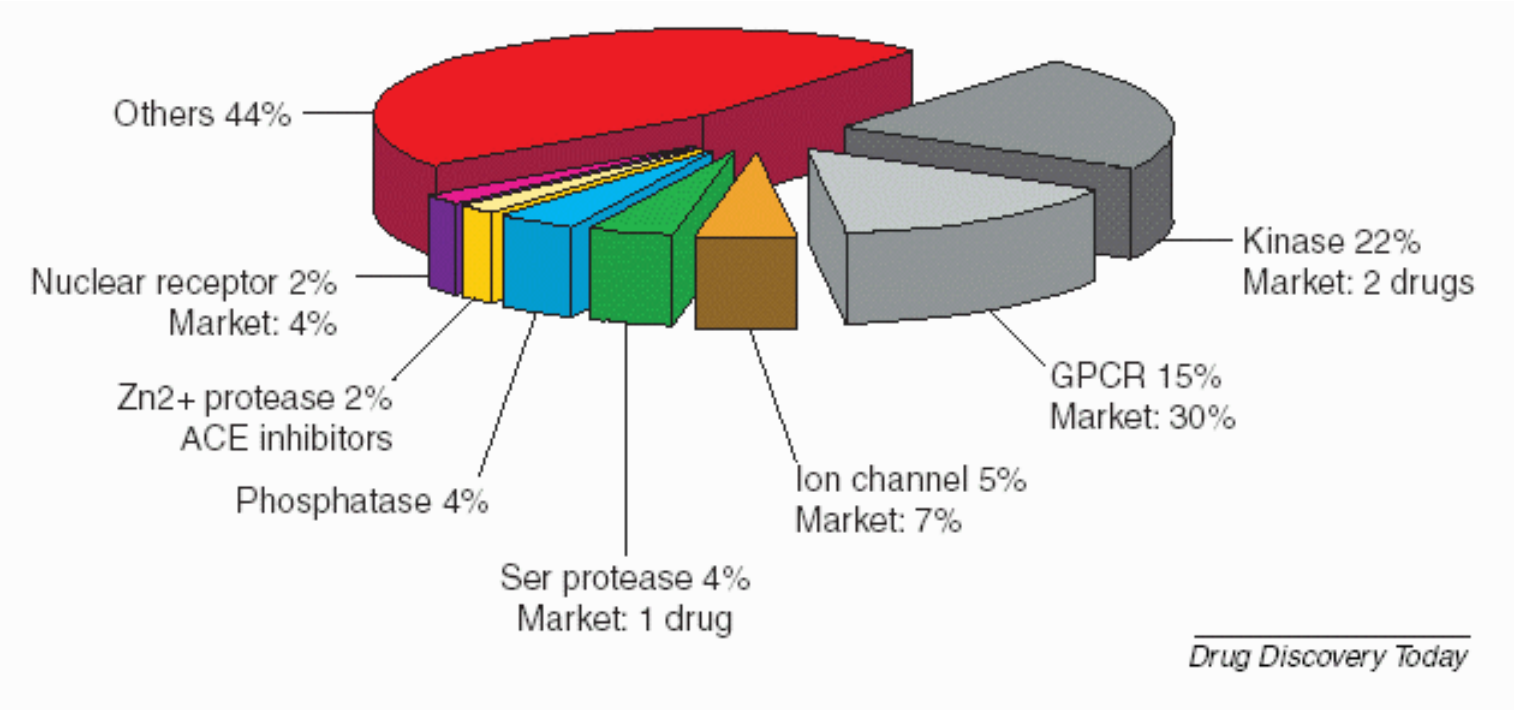
data: Hopkins & Groom, *Nat.Rev.Drug.Disc.* **1** (2002) 727

Enzymatic targets



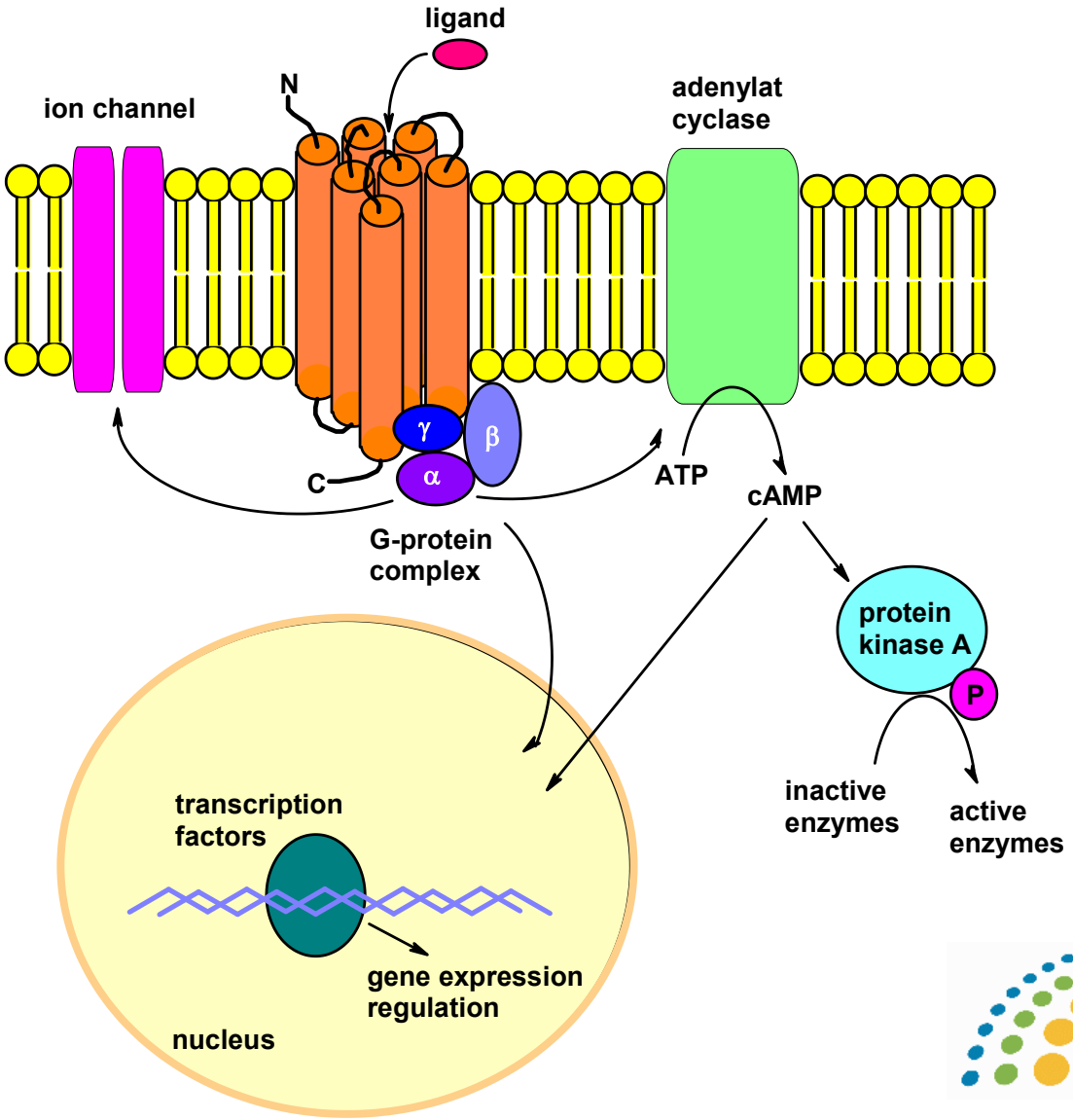
Distribution within the class of enzymes

typical targets



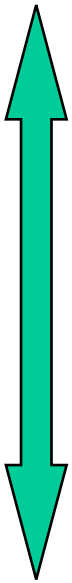
contribution to the human genome and marketed drugs
about 500 enzymes have been used as targets
100,000 estimated potential targets in the genome

GPCRs and other targets



How do drugs interact with targets ?

proteome



genome

enzymes: substrate analogs, competitive ligands, reversible and irreversible inhibitors, allosteric modulators, protein-protein inhibitors

receptors: antagonists and agonists, orthosteric and allosteric ligands.

ion channels: openers and blockers (inhibitors)

transporters: inhibitors, e.g for (re-)uptake

nuclear receptors: binding to specific DNA-motives

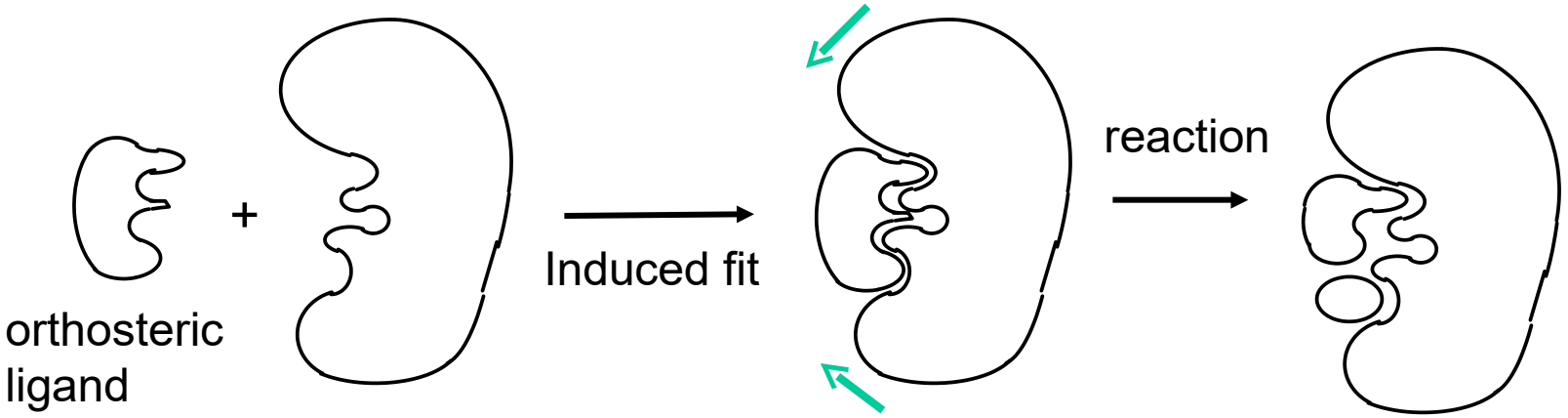
DNA: binding to groves, intercalation, alkylation, etc.

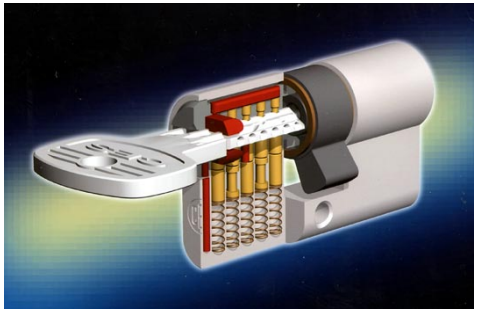
→ Possible targets can be found in a multitude of cell compartments and at different loci.

Problems: Drug delivery and drug transport / distribution

Drugs: mode of action (I)

Normal enzymatic turn-over



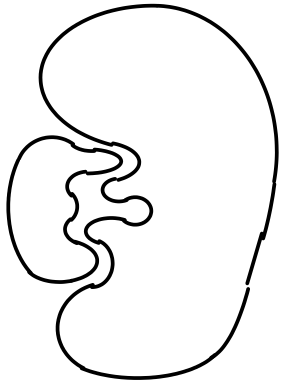


lock and key principle

Drugs: mode of action (II)

competitive inhibitor:

higher affinity than natural substrate, directly acting at the orthosteric site



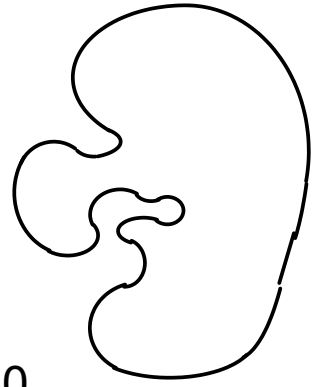
allosteric inhibitor/effector:

prevents binding by modifying the conformation, but at the allosteric site

Irreversible binding:

chemical reaction leads to inactivation of the enzyme

e.g. acetyl-salicylic acid acetylates Ser530 of Cyclooxygenase, ibrutinib reacts with Cys481 of Bruton‘s tyrosine kinase



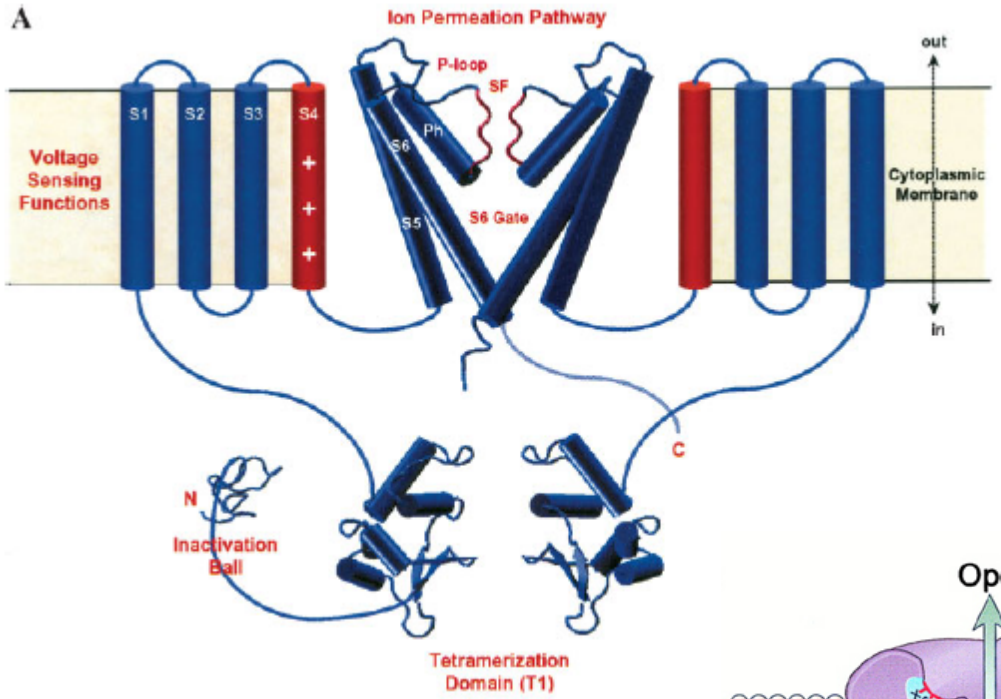
Anti-metabolite:

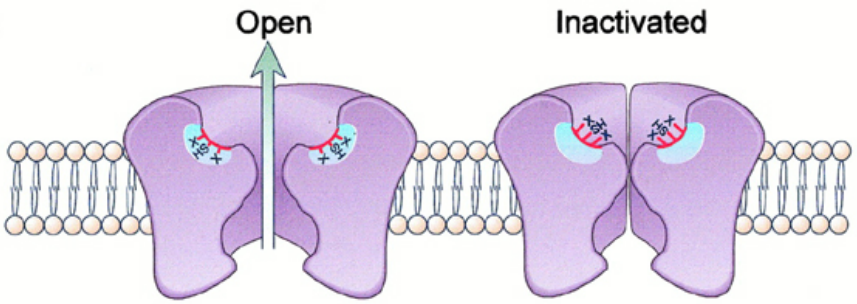
Competitive alternate („wrong“) substrate

e.g. methotrexate instead of dihydrofolate, antiviral nucleoside analoges

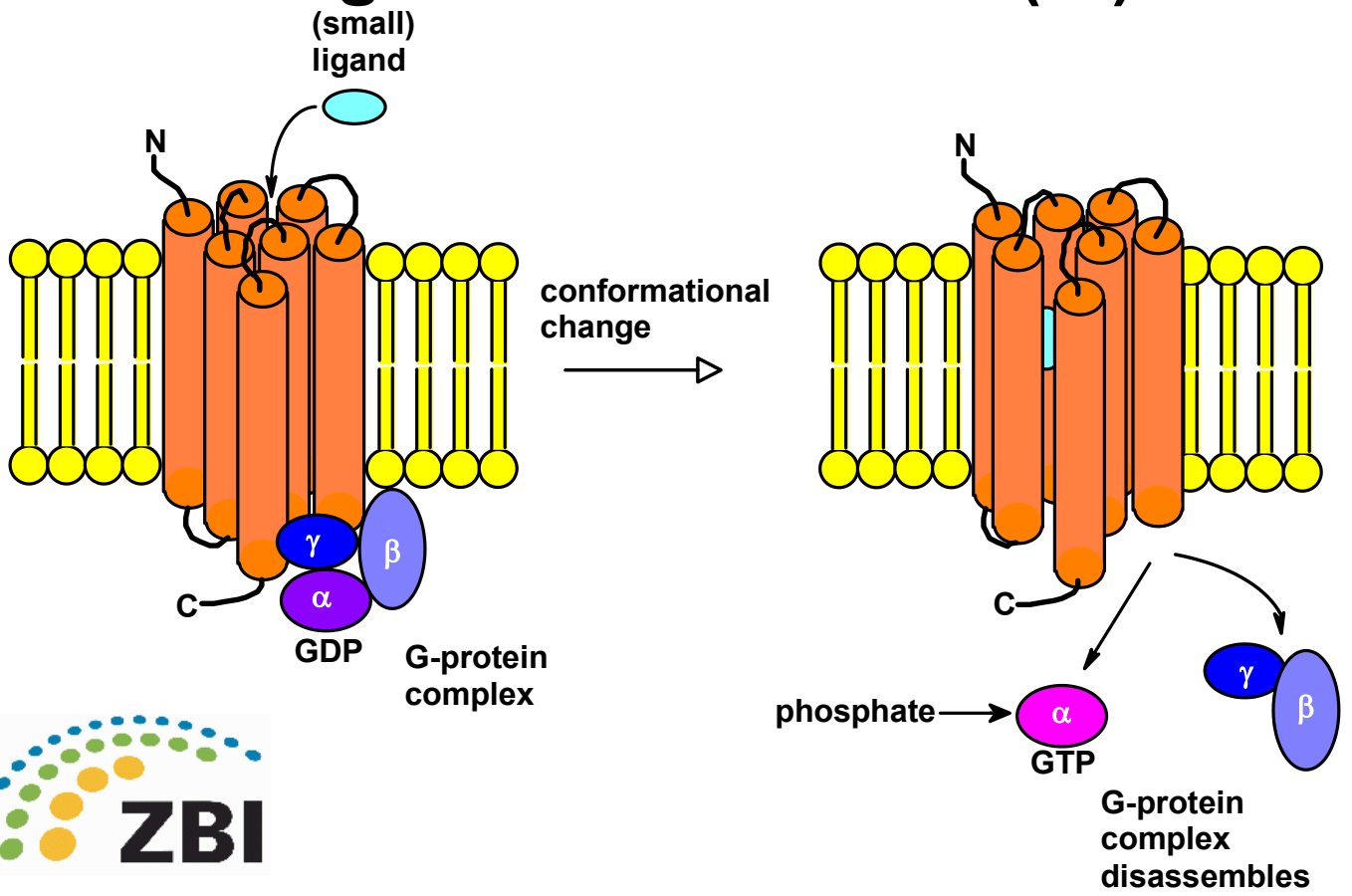
Drugs: mode of action (III)

Ion channels: Mode of action by ligand binding, and also indirectly through receptors





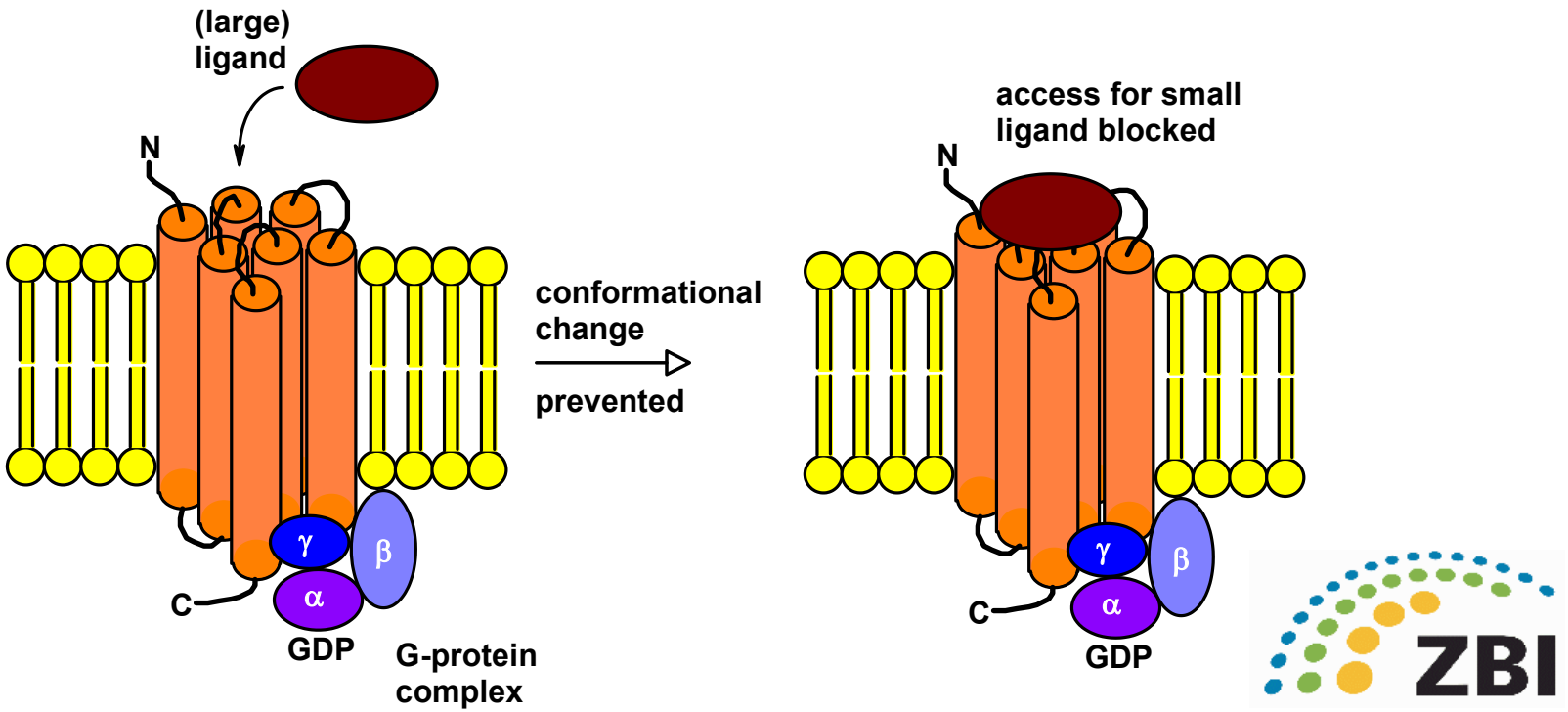
Drugs: mode of action (IV)



agonist: (orthosteric) ligand that causes an intrinsic effect (response of the receptor)

partial agonist: weakly working agonist with high binding affinity, thus also working as antagonist

Drugs: mode of action (V)

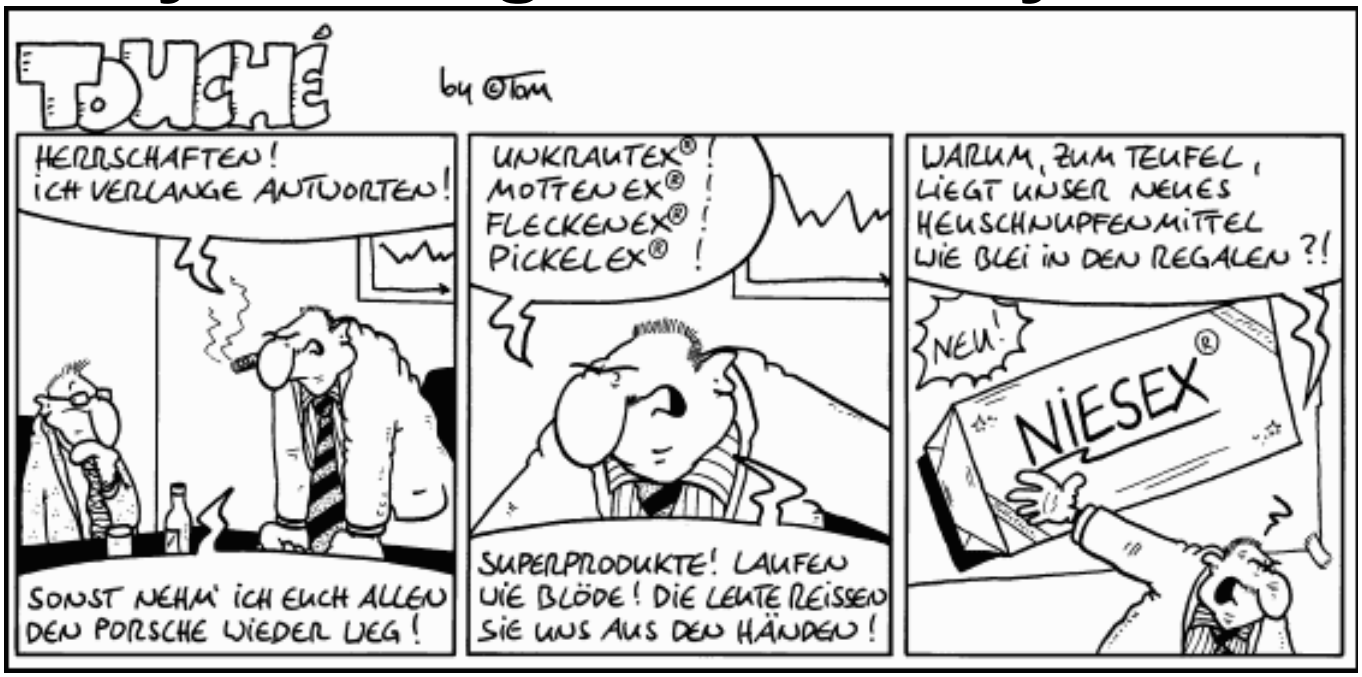


antagonist: ligand that prevents binding of the agonist, either directly (competitive binding at the orthosteric site) or indirectly (at allosteric site, prevents adoption of the reactive conformation)

inverse agonist: ligand stabilizing the inactive conformation

functional antagonist: prevents receptor response by a different mode of action

Why do drugs have funny names ?



Examples for such faults in naming products exist !

Nasonex®
(Mometasone Furoate Monohydrate Aqueous Nasal Spray)

Naming of drugs (I)

The **trade name** of a drug is usually chosen very carefully. Associative and speach-psychological aspects are considered.

Example within the german language:

The more x and y are appearing in the name, the more toxic.

Acetylsalicylsäure → Aspirin®

Problems will occur, if a product should get the same name throughout all countries. Examples:

Twix® (earlier: Raider)

Naming of drugs (II)

Furthermore, legal aspects have to be considered:
existing words and words that imply a direct connection or target a specific consumer group cannot be protected.

Example: „Schülerschokolade“ is not possible in Germany

Thus a lot of inspiration is required to find a pleasant sounding name. Frequently syllables and foreign words (latin, greek, spanish) are used that bear associations.

c.f. names for cars

® this name is approved and protected.

™ the producer indicates his intention to have this name protected.

Naming of drugs (III)

For the naming of the actual chemical substances there are also some (loose and empirical) guidelines.

Such names are adopted as „International Nonproprierary Name“ (INN) or „United States Adopted Name“ (USAN) at the lastest upon patent application.

Most of the time, the therapeutic class can be identified solely by the name. (similar names for substances with similar function.)

Prefixes and suffixes reflect chemical modification of the root compound.

Examples: ibufenac, clofenac, diclofenac, oxidanac

→ (unselective) Cyclooxygenase inhibitors

Naming of drugs (IV)

The World Health Organization (WHO) publishes updates regarding the use of stems in the selection of International Nonproprietary Names (INN) for pharmaceutical substances

Example: all drugs carrying the suffix (=stem) –coxib are selective Cycloxygenase II inhibitors:

celecoxib, cimicoxib, deracoxib, etoricoxib, firocoxib,

lumiracoxib, mavacoxib, parecoxib, robenacoxib, rofecoxib,

tilmacoxib, valdecoxib

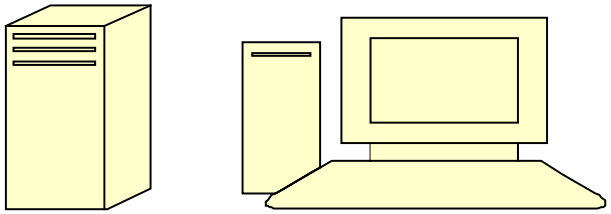
In such cases the drug target is obvious.

Compound Databases

Size of typical substance libraries (2022)

ACD	12,000,000 chemicals	commercial
World Drug Index	80,000 compounds	
USAN	<10,000 in clinical trials	
virtual library	≈100,000 compounds	company, in house

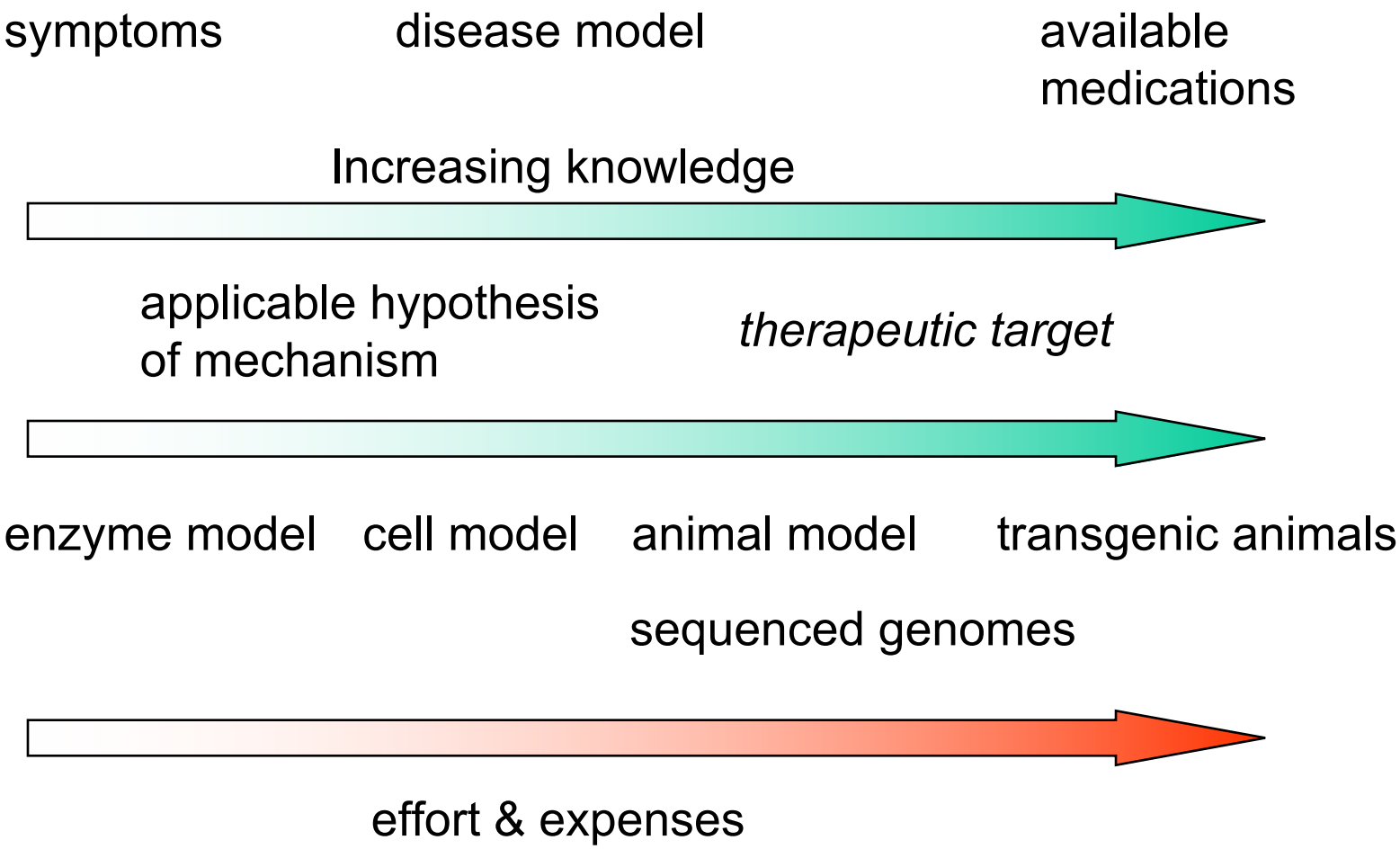
PubChem	> 112,000,000 compounds	NCBI	academic
ChEMBL	> 2,200,000 compounds	EMBL	
DrugBank	> 500,000 drugs	Uni. Alberta	
ZINC15	>750,000,000 compounds	UCSF	



Investment per new chemical entity: >800,000 $

New chemical entities per year: ca. 15 strongly fluctuating

towards the drug (I)



Evolution of Disease Symptoms with Time

1800 1900 2000

Disease causing agents
lack of hygiene
germs, bacteria
viruses

enviromental
influence
carcinogens

life style

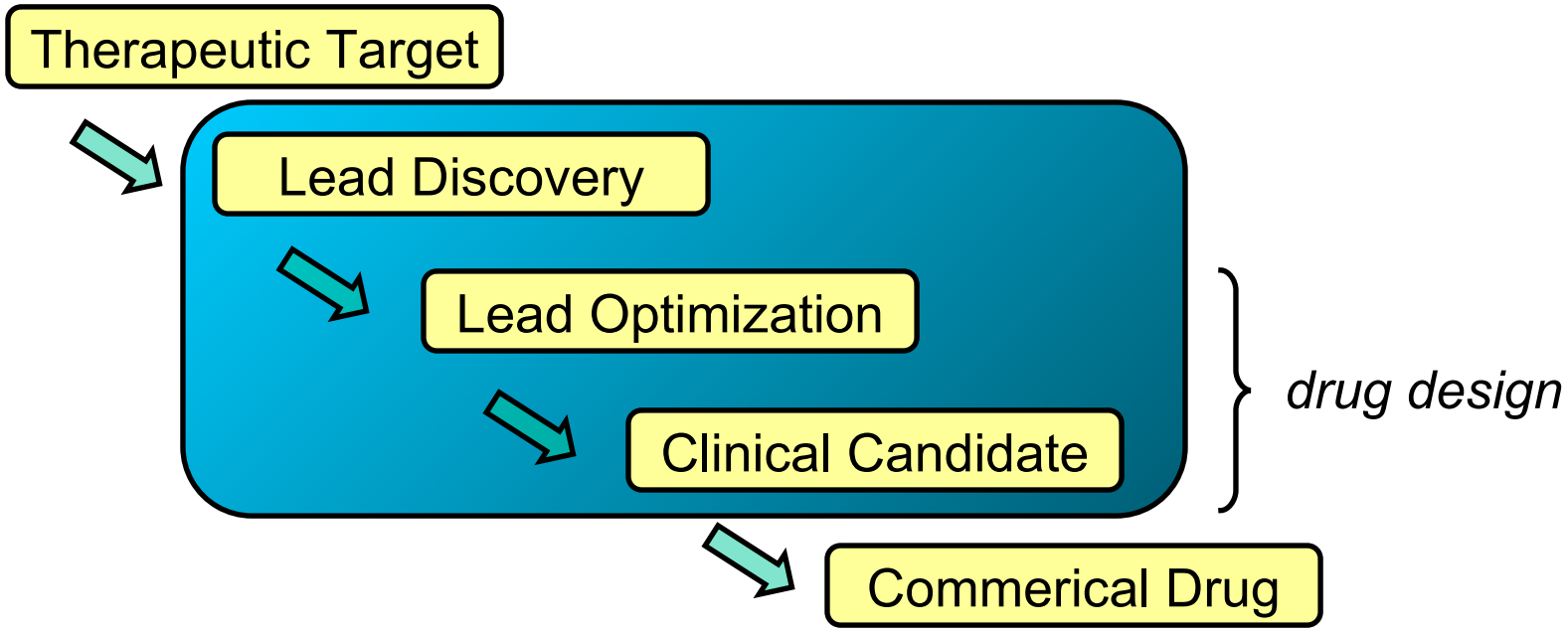
genetic
disposition
susceptibility

bioethic component

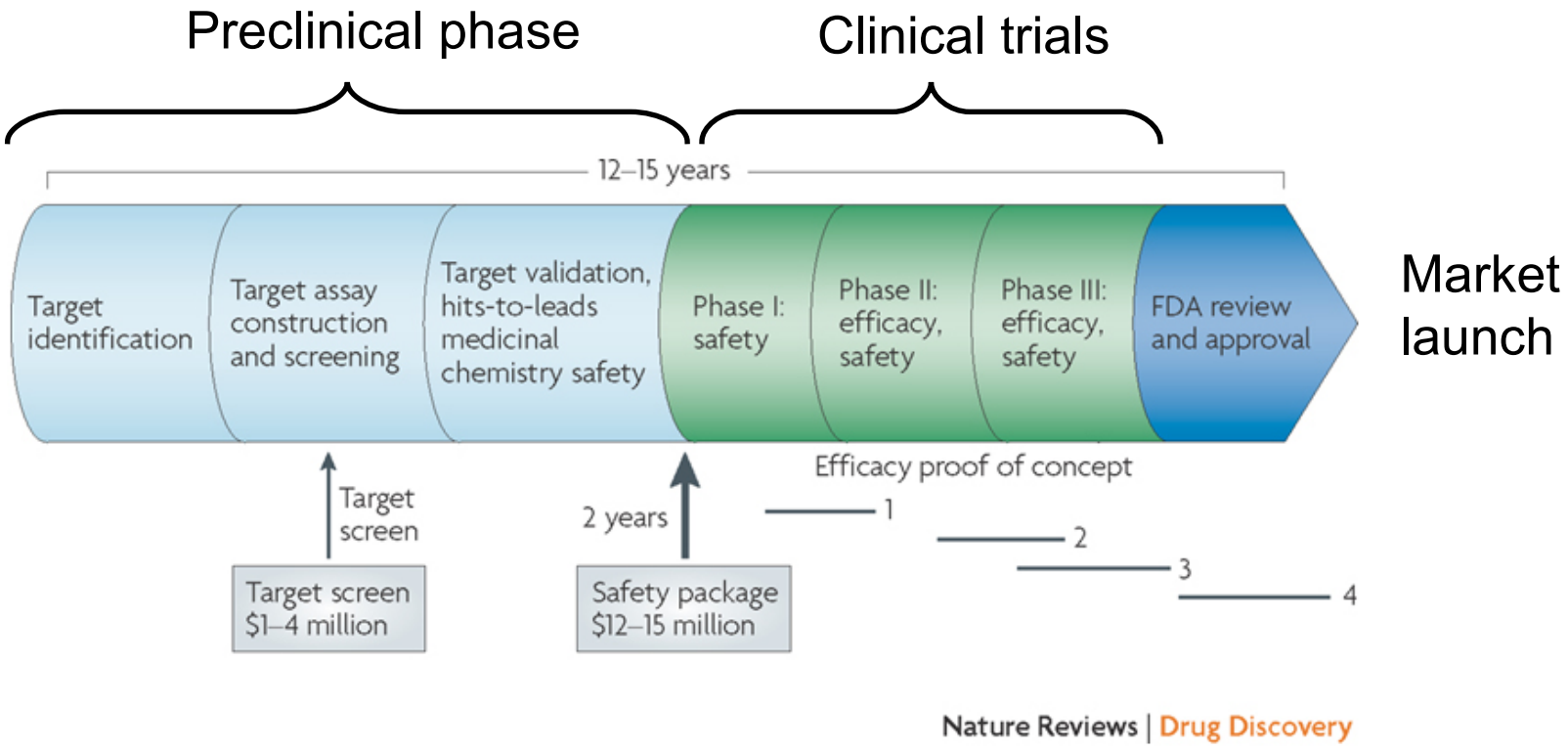
accepted legal definition of diseases

legal regulation for drug marketing (e.g. by the FDA)

The preclinical phase



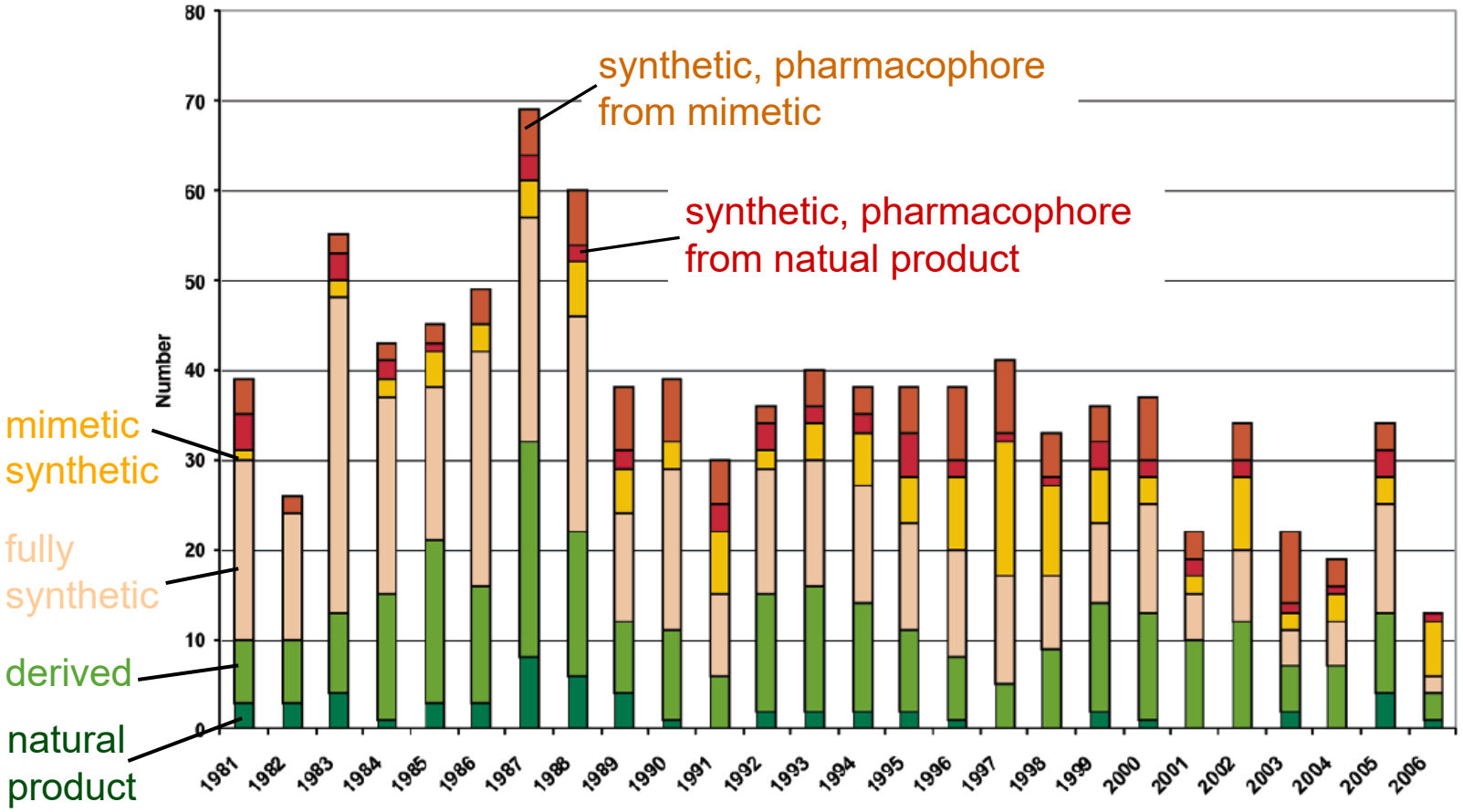
The drug discovery pipeline



A.D. Roses *Nature Reviews Drug Discovery* **7** (2008) 807.

Trend in approving new drugs

Drugs approved by the FDA (1981 - 2006)



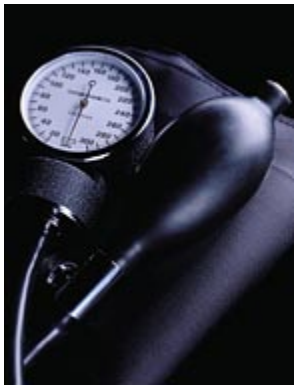
Lit: D.J. Newman, G.M. Cragg *J.Nat.Prod.* **70** (2007) 461-477.

towards the drug (II)

Example: arterial hypertension

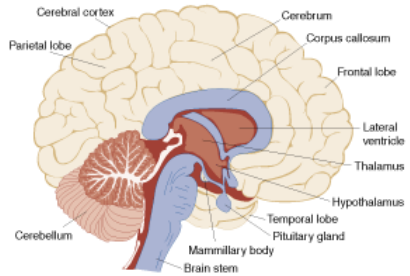
Arterial hypertension [Arterielle Hypertonie] is a frequently observed condition (about 10 - 25% of all adults are affected). Persistent hypertension can lead to damage of blood vessels, the eyes, and the kidneys. → symptoms

category	systolic		diastolic	
optimum	<120	and	<80	
normal	<130	and	<85	
normal-high	130 - 139	or	85 - 89	
mild HD	140 - 159	or	90 - 99	
moderate HD	160 - 179	or	100 - 109	
strong HD	>180	or	>110	mm (Hg)



source: *Archives Int. Med.* **157** (1997) 2413.

Regulation of the blood pressure (simplyfied)



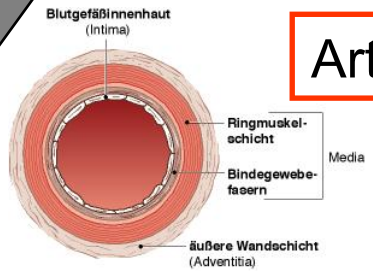
blood volume

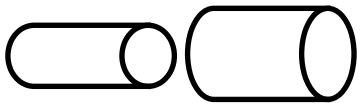
hormones

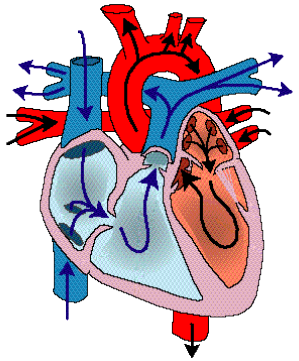
sympathicus ↑

parasympathicus ↓

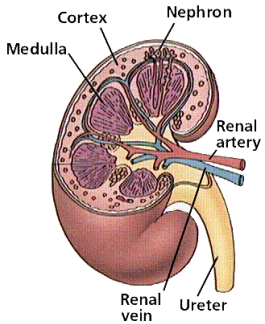
Arterial blood pressure

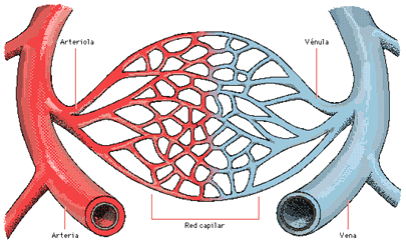






heart
(pumping)
capacity





peripheral resistance

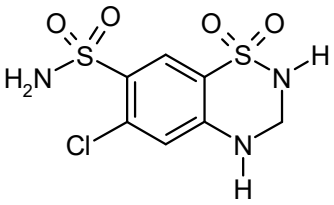
salt deposits Na^+, K^+, Ca^{2+}

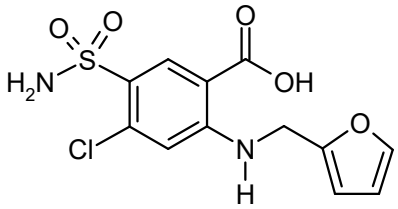
diuretica and saluertica

Ions in the blood and in other salt deposits bind water. By elimination of these ions the volume of the blood can be reduced.

This effect is caused by diuretica and saluertica:

Examples: hydrochlorothiazide, furosemide

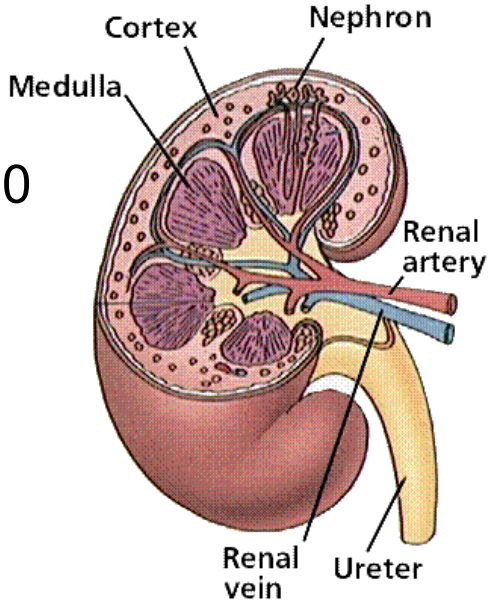




Therapeutic administration of thiazides since 1960

Disadvantages / side effects:

deficiency of potassium
increased level of uric acid [Harnsäure]
increased level of fatty acids in the serum
not suitable with diabetes

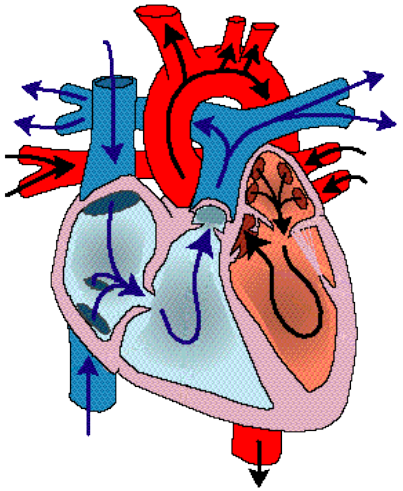


α and β-blocker

Act relaxing via the peripheral nervous system and reduce the pumping capacity of the heart.

Examples: prasozin, tetrazosin, doxazosin, propanolol, atenolol, labetalol, pindolol

Simultaneously, the hormonal control is affected, whereby the peripheral resistance is diminished.



Therapeutic administration since 1970

Disadvantages and side effects:

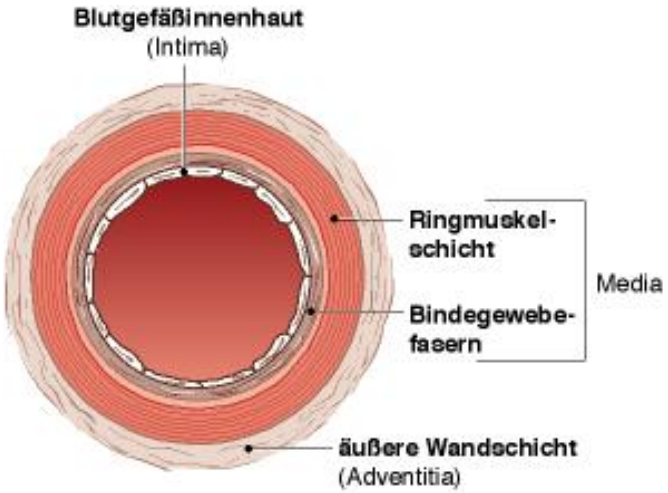
withdrawl symptomes
reduced capacity of the heart [Herzinsuffizienz]
increased levels of fatty acids in the serum
effects on the central nervous system

vasodilators and calcium antagonists

Act relaxing on the smooth muscles of the arterias and thereby reduce the resistance.

Bind to the *h*AT_2-receptor or inhibit the calcium pump

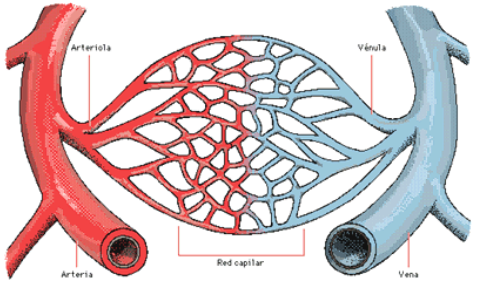
Examples: hydralazine, minoxidil, diazoxide, verapamil, diltiazem, nifedipine



Therapeutic administration since 1980

Disadvantages and side effects:

Predominately on the function of the heart



Angiotensin Coverting Enzyme Inhibitors

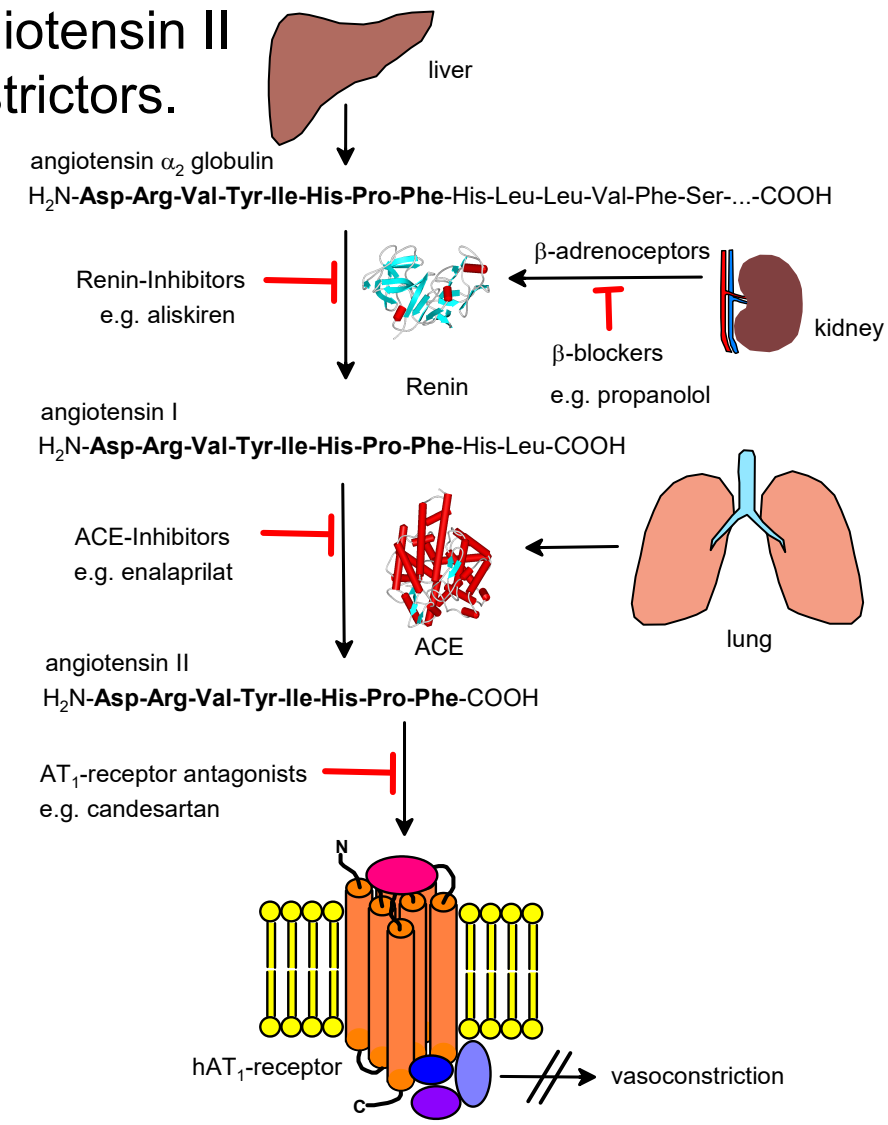
The endogenic oligopeptide Angiotensin II is one of the strongest vasoconstrictors. By inhibiting the angiotensin converting enzyme (ACE), the synthesis of Angiotensin II is disabled.

Examples: captopril, fosinopril, quinapril, enalapril

Therapeutic administration since 1990

disadvantages:

fetotoxic (pregnancy)



Angiotensin-II antagonists

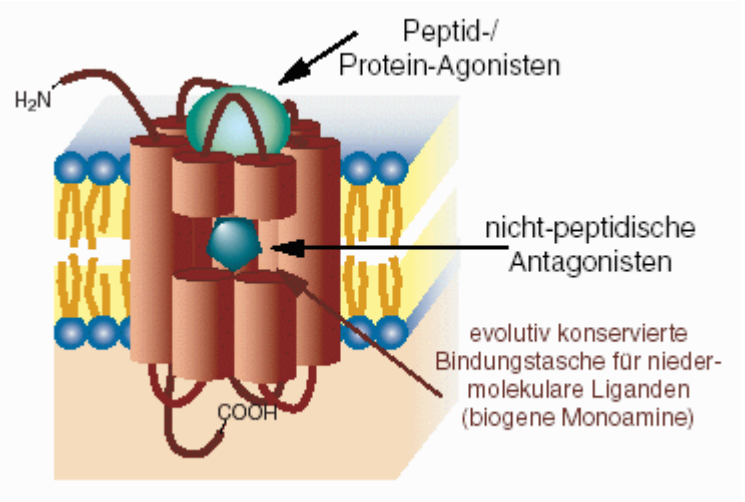
competitive binding of non-peptidic compounds to the *h*AT_1-receptor (GPCR), which is the binding site of Angiotensin II.

Examples: losartan, valsartan, irbesartan, candesartan, telmisartan, olmesartan, forsartan

therapeutic administration since 1995

disadvantages:

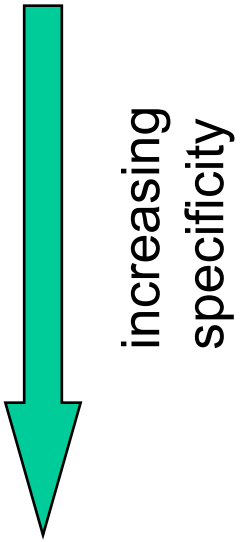
same as for ACE-inhibitors



Picture source: M. Gurrath *Pharm. i. u. Zeit* **288** (2001) 288.

„Evolution“ of hypertension targets over time

targets	therapeutic class
kidney	diuretica, saluretica
nervous system	α and β-blockers
calcium channels (adrenal gland)	calcium channel blockers
hAT_2-receptor	vasodilators
ACE	ACE-inhibitors
hAT_1-receptor	Angiotensin II antagonists



Methods for Determining Atomic Structures

X-ray and electron microscopy

Diffraction and scattering of electromagnetic waves, respectively electrons

Pro: high resolution

Con: „artificial“ conditions
solid state, crystal

NMR (nuclear magnetic resonance)

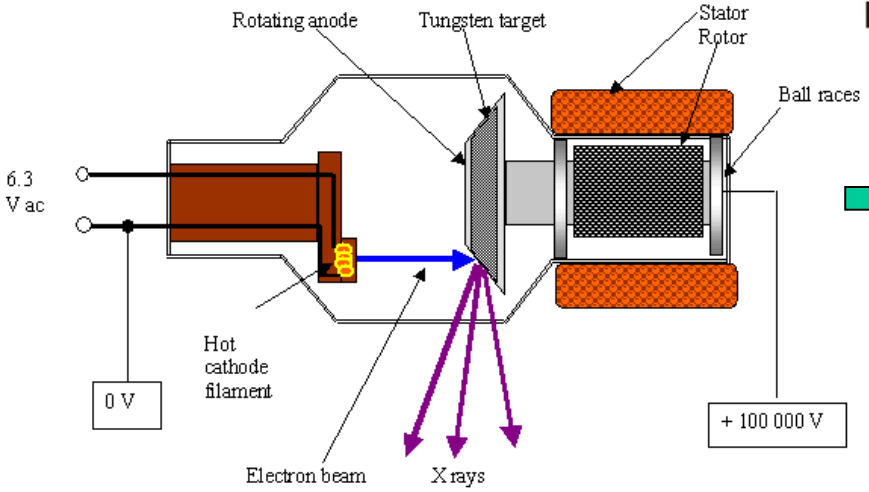
Absorption of electromagnetic waves

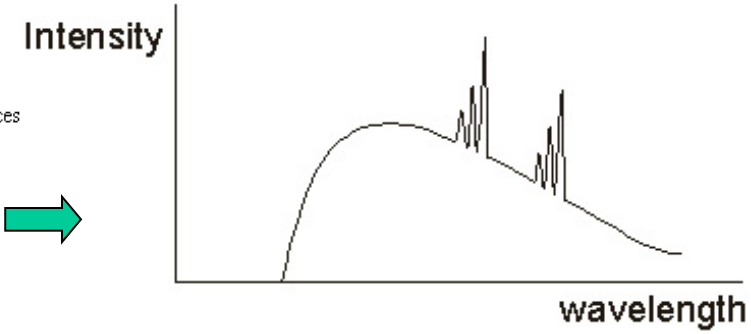
Pro: „natural“ conditions
aqueous solution

Con: only for small proteins
<100 amino acids

Obtaining X-Ray structures (I)

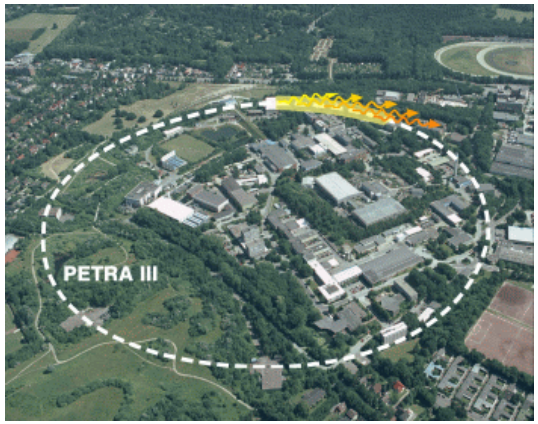
Conventional X-Ray scattering

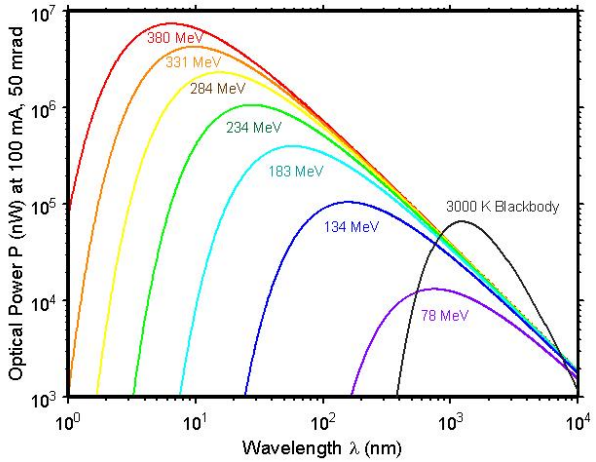




Uneven distribution of wave lengths: characteristic intensity peaks

Synchrotron scattering

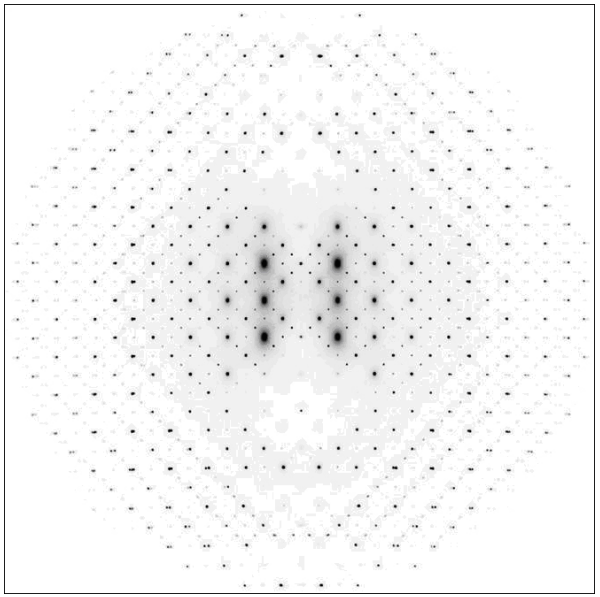




Continuous distribution of wave lengths

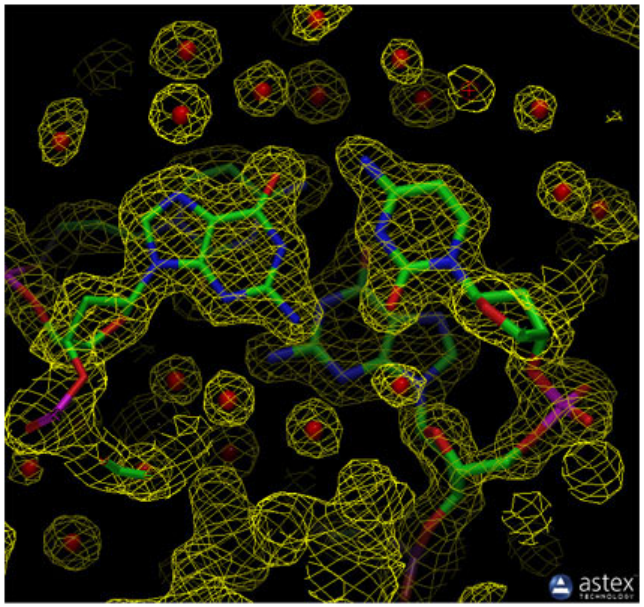
Obtaining X-Ray structures (II)

The arrangement of atoms in the crystal gives rise to a diffraction pattern



Mathematical transformation &

Idea how the actual structure looks like



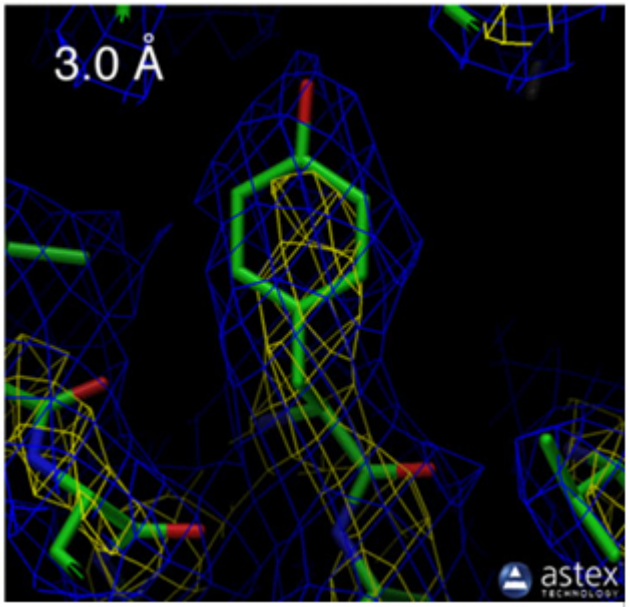
Electron density map

3D coordinates can be obtained from www.rcsb.org (Protein Data Bank) and electron density maps from http://www.ebi.ac.uk/pdbe/

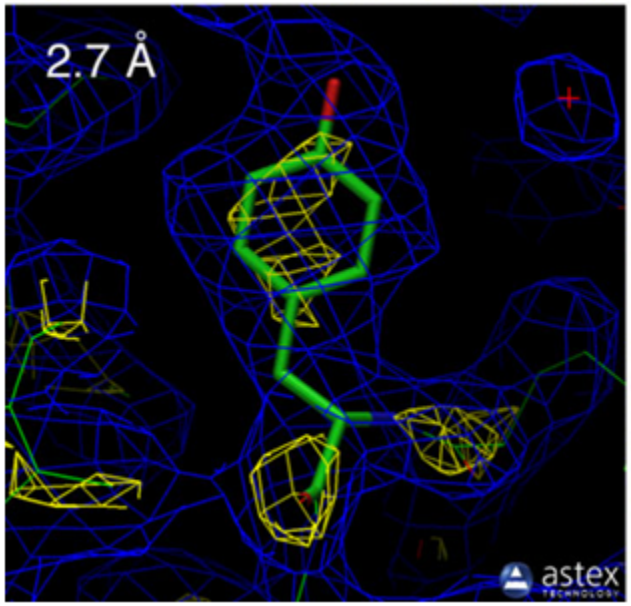
Accurracy of X-Ray structures (I)

The resolution given in .pdb files tells us the smallest wavelength (in Å) the crystal was able to diffract.

→ We “see” objects of that size



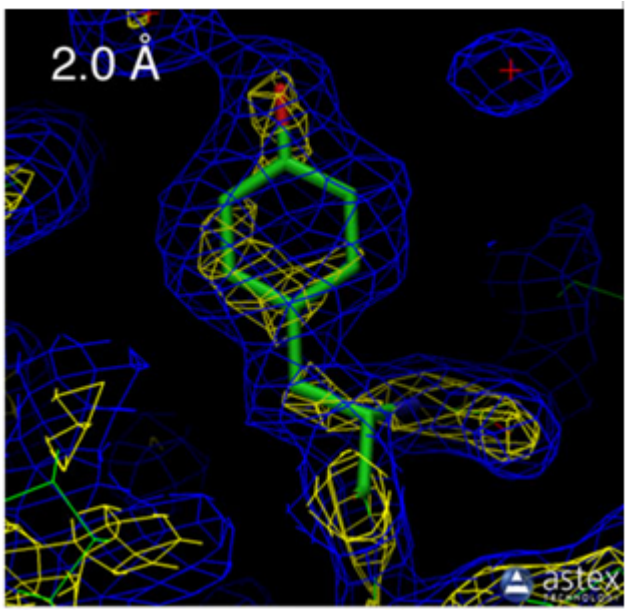
Backbone and putative positions of side chains

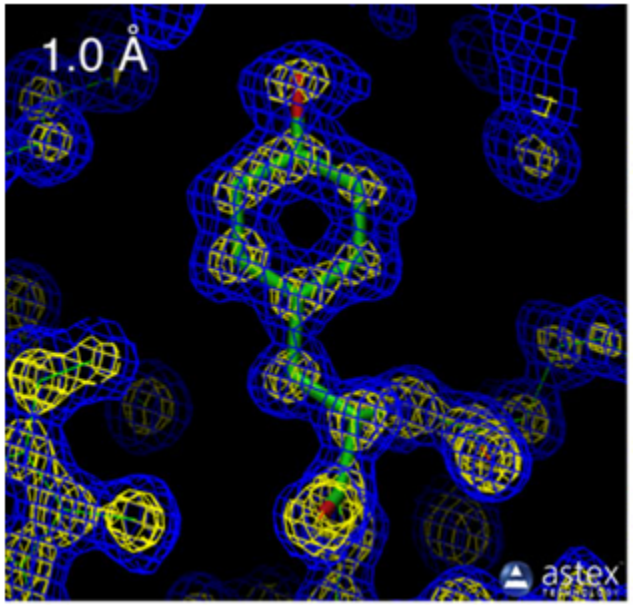


Distinguishing large and small side chains

Accurracy of X-Ray structures (II)

The atomic coordinates are fitted into the electron density grid using a force field





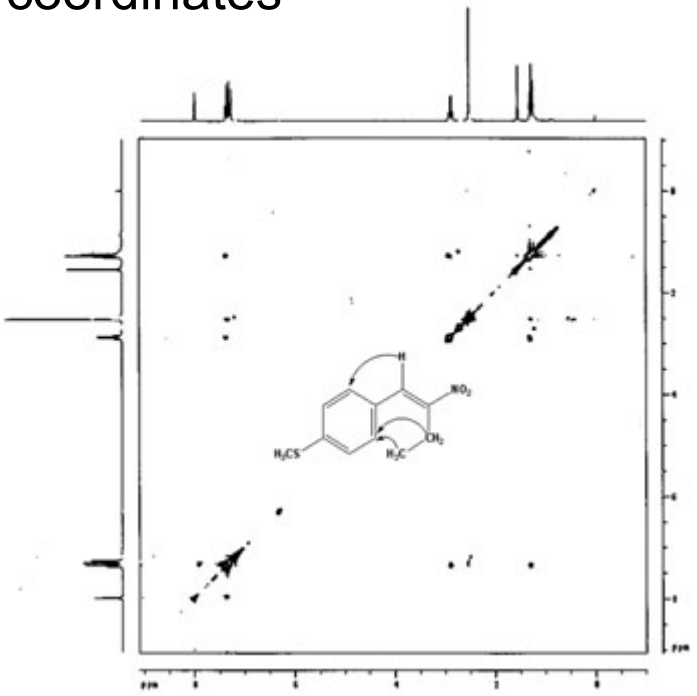
Different rotamers of the side chains can be assigned

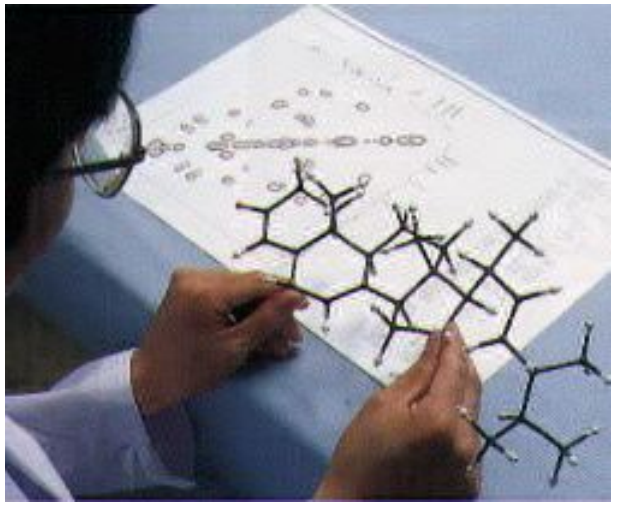
Atomic coordinates can be assigned unambigously

NMR Structures (I)

The distance between atoms can be derived from the intensities of the cross-peaks in the 2D-NMR spectrum. Like in X-ray scattering an idea how the actual structure looks like is required.

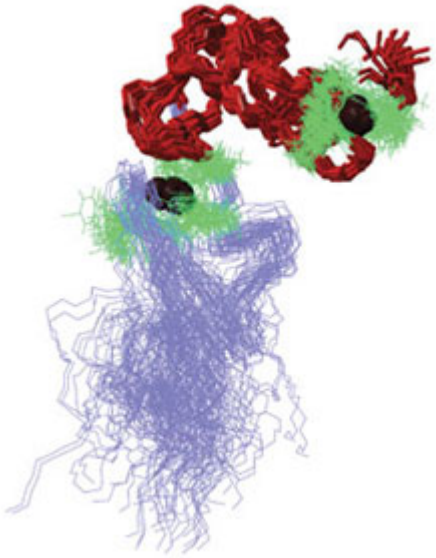
→ constraint force field optimization of the atomic coordinates





NMR Structures (II)

Due to the dynamic behavior in solution and limited time resolution of the NMR, an „averaged“ structure is obtained.



Such .pdb files usually contain 10 individual solutions (models) and the resulting averaged structure.