## **Typical diseases**

The search for pharmaceutical drugs used to be rather straight forward until recent times:

A wealth of information about the disease, its causes, and the clinical symptoms were readily available. Thus the starting point for the pharmacological therapy was known.

Example: inhibition of an enzyme

Thus the target was fixed. Frequently, experience with existing medications was available. Therefore a *valid target* or at least a *drugable target* was present.

 $\rightarrow$  The *target* undergoes a change of its activity caused by the drug

Often, multiple targets can be exploited to treat the same disease.  $\rightarrow$  Therapeutic Categories



## **Therapeutic Categories**

Grouping drugs under the aspect of their pharmacological and therapeutic application results in about 200 categories:

ACE Inhibitor **Analgesic, Dental** Adrenocortical Suppressant Analgesic, Narcotic Adrenocorticotropic Hormones Analgesic, Non-narcotic Aldose Reductase Inhibitors **Androgens** Aldosterone Antagonists Anesthetics, Inhaled  $\alpha$ -adrenergic Agonists  $\alpha$ -adrenergic Agonists  $\alpha$ -adrenergic Blockers **Anesthetics, Local**  $\alpha$ -Glucosidase Inhibitors **Angiotensin II Antagonists** Anabolic Streroids **Anabolic Streroids** 



see the corresponding section in the Merck Index



In most cases it is not obvious to recognize the treated disease from a therapeutic class (at least for non-medical persons), sometimes not even the actual molecular target(s).

## **Flow of information in a drug discovery pipeline**



Drug Discovery Today

## **typical targets**



Fractional content of marketed drugs according to their biochemical targets

data: Hopkins & Groom, *Nat.Rev.Drug.Disc.* **1** (2002) 727



2nd Lecture **Modern Methods in Drug Discovery WS24/25** 4

#### **Enzymatic targets**



Distribution within the class of enzymes



### **typical targets**



contribution to the human genome and marketed drugs about 500 enzymes have been used as targets 100,000 estimated potential targets in the genome



### **GPCRs and other targets**



2nd Lecture **Modern Methods in Drug Discovery WS24/25** 7

## **How do drugs interact with targets ?**

proteome

genome

enzymes: substrate analogs, competitive ligands, reversible and irreversible inhibitors, allosteric modulators, protein-protein inhibitors

receptors: antagonists and agonists, orthosteric and allosteric ligands.

ion channels: openers and blockers (inhibitors)

transporters: inhibitors, e.g for (re-)uptake

nuclear receptors: binding to specific DNA-motives

DNA: binding to groves, intercalation, alkylation, etc.

 $\rightarrow$  Possible targets can be found in a multitude of cell compartments and at different loci.

Problems: Drug delivery and drug transport / distribution

### **Drugs: mode of action (I)**

#### Normal enzymatic turn-over





## **Drugs: mode of action (II)**

competitive inhibitor:

higher affinity than natural substrate, directly acting at the orthosteric site



Irreversible binding:

chemical reaction leads to inactivation of the enzyme

e.g. acetyl-salicylic acid acetylates Ser530 of Cyclooxygenase, ibrutinib reacts with Cys481 of Bruton's tyrosine kinase

allosteric inhibitor/effector:

prevents binding by modifying the conformation, but at the allosteric site



Anti-metabolite:

Competitive alternate ("wrong") substrate

e.g. methotrexate instead of dihydrofolate, antiviral nucleoside analoges



## **Drugs: mode of action (III)**

Ion channels: Mode of action by ligand binding, and also indirectly through receptors





**agonist**: (orthosteric) ligand that causes an intrinsic effect (response of the receptor)

**partial agonist**: weakly working agonist with high binding affinity, thus also working as antagonist

## **Drugs: mode of action (V)**



**antagonist**: ligand that prevents binding of the agonist, either directly (competitive binding at the orthosteric site) or indirectly (at allosteric site, prevents adoption of the reactive conformation)

**inverse agonist**: ligand stabilizing the inactive conformation

**functional antagonist**: prevents receptor response by a different mode of action

2nd Lecture **120 Modern Methods in Drug Discovery WS24/25** 13

## **Why do drugs have funny names ?**



Examples for such faults in naming products exist !





# **Naming of drugs (I)**

The **trade name** of a drug is usually chosen very carefully. Associative and speach-psychological aspects are considered.

Example within the german language:

The more x and y are appearing in the name, the more toxic.

Acetylsalicylsäure  $\rightarrow$  Aspirin®

Problems will occur, if a product should get the same name throughout all countries. Examples:

Twix® (earlier: Raider)





# **Naming of drugs (II)**

Furthermore, legal aspects have to be considered: existing words and words that imply a direct connection or target a specific consumer group cannot be protected.

Example: "Schülerschokolade" is not possible in Germany

Thus a lot of inspiration is required to find a pleasant sounding name. Frequently syllables and foreign words (latin, greek, spanish) are used that bear associations.

c.f. names for cars





® this name is approved and protected.

™ the producer indicates his intention to have this name protected.

2nd Lecture **120 Modern Methods in Drug Discovery WS24/25** 2nd Lecture 16

# **Naming of drugs (III)**

For the naming of the actual chemical substances there are also some (loose and empirical) guidelines.

Such names are adopted as "International Nonproprierary Name" (INN) or "United States Adopted Name" (USAN) at the lastest upon patent application.

Most of the time, the therapeutic class can be identified solely by the name. (similar names for substances with similar function.)

Prefixes and suffixes reflect chemical modification of the root compound.

Examples: ibufenac, clofenac, diclofenac, oxidanac

 $\rightarrow$  (unselective) Cyclooxygenase inhibitors

# **Naming of drugs (IV)**

The World Health Organization (WHO) publishes updates regarding the use of stems in the selection of International Nonproprietary Names (INN) for pharmaceutical substances

Example: all drugs carrying the suffix (=stem) –coxib are selective Cycloxygenase II inhibitors:

celecoxib, cimicoxib, deracoxib, etoricoxib, firocoxib,

lumiracoxib, mavacoxib, parecoxib, robenacoxib, rofecoxib,

tilmacoxib, valdecoxib

In such cases the drug target is obvious.





## **Compound Databases**

#### **Size of typical substance libraries (2022)**

ACD 12,000,000 chemicals World Drug Index 80,000 compounds USAN <10,000 in clinical trials virtual library ≈100,000 compounds commercial company, in house

PubChem > 112,000,000 compounds NCBI ChEMBL > 2,200,000 compounds EMBL

DrugBank > 500,000 drugs Uni. Alberta

ZINC15 >750,000,000 compounds UCSF

Investment per new chemical entity: >800,000 \$ New chemical entities per year: ca. 15 strongly fluctuating

2nd Lecture **19** Nodern Methods in Drug Discovery WS24/25

academic

## **towards the drug (I)**



## **Evolution of Disease Symptoms with Time**



## **The preclinical phase**



## **The drug discovery pipeline**



Nature Reviews | Drug Discovery

#### A.D. Roses *Nature Reviews Drug Discovery* **7** (2008) 807.

## **Trend in approving new drugs (I)**

Drugs approved by the FDA (1981 - 2006)



Lit: D.J. Newman, G.M. Cragg *J.Nat.Prod.* **70** (2007) 461-477.

## **Trend in approving new drugs (II)**

#### FDA drug approvals by year

small molecule

biologic or other



## **towards the drug (II)**

#### **Example: arterial hypertension**

Arterial hypertension [Arterielle Hypertonie] is a frequently observed condition (about 10 - 25% of all adults are affected). Persistent hypertension can lead to damage of blood vessels, the eyes, and the kidneys.  $\rightarrow$  symptoms





source: *Archives Int. Med.* **157** (1997) 2413.

#### **Regulation of the blood pressure (simplyfied)**



### **diuretica and saluertica**

Ions in the blood and in other salt deposits bind water. By elimination of these ions the volume of the blood can be reduced.

This effect is caused by diuretica and saluertica:

Examples: hydrochlorothiazide, furosemide



Renal Ureter vein

not suitable with diabetes

### **and -blocker**

Act relaxing via the peripheral nervous system and reduce the pumping capacity of the heart.

Examples: prasozin, tetrazosin, doxazosin, propanolol, atenolol, labetalol, pindolol

Simultaneously, the hormonal control is affected, whereby the peripheral resistance is diminished.

Therapeutic administration since 1970

Disadvantages and side effects:

withdrawl symptomes reduced capacity of the heart [Herzinsuffizienz] increased levels of fatty acids in the serum effects on the central nervous system



#### **vasodilators and calcium antagonists**

Act relaxing on the smooth muscles of the arterias and thereby reduce the resistance.

Bind to the *h*AT<sub>2</sub>-receptor (vasodilators) or inhibit the calcium pump (calcium antagonists)

Examples: hydralazine, minoxidil, diazoxide, verapamil, diltiazem, nifedipine

Therapeutic administration since 1980

Disadvantages and side effects:

Predominately on the function of the heart





## **Angiotensin Coverting Enzyme Inhibitors**

The endogenic oligopeptide Angiotensin II is one of the strongest vasoconstrictors.

By inhibiting the angiotensin converting enzyme (ACE), the synthesis of Angiotensin II is disabled.

Examples: captopril, fosinopril, quinapril, enalapril

Therapeutic administration since 1990

disadvantages:

fetotoxic (pregnancy)



## **Angiotensin-II antagonists**

competitive binding of non-peptidic compounds to the *h*AT<sub>1</sub>receptor (GPCR), which is the binding site of Angiotensin II.

Examples: losartan, valsartan, irbesartan, candesartan, telmisartan, olmesartan, forsartan

therapeutic administration since 1995

disadvantages:

same as for ACE-inhibitors



Picture source: M. Gurrath *Pharm. i. u. Zeit* **288** (2001) 288.

## **"Evolution" of hypertension targets over time**





## **Methods for Determining Atomic Structures**

#### **X-ray and electron microscopy**

Diffraction and scattering of electromagnetic waves, respectively electrons

Pro: high resolution Con: "artificial" conditions solid state, crystal

**NMR** (nuclear magnetic resonance)

Absorption of electromagnetic waves

Pro: "natural" conditions Con: only for small proteins aqueous solution <100 amino acids

## **Obtaining X-Ray structures (I)**



## **Obtaining X-Ray structures (II)**

The arrangement of atoms in the crystal gives rise to a diffraction pattern



#### Electron density map

3D coordinates can be obtained from www.rcsb.org (Protein Data Bank) and electron density maps from http://www.ebi.ac.uk/pdbe/

## **Accurracy of X-Ray structures (I)**

The resolution given in .pdb files tells us the smallest wavelength (in Å) the crystal was able to diffract.

 $\rightarrow$  We "see" objects of that size



## **Accurracy of X-Ray structures (II)**

The atomic coordinates are fitted into the electron density grid using a force field





Different rotamers of the side chains can be assigned

Atomic coordinates can be assigned unambigously

## **NMR Structures (I)**

The distance between atoms can be derived from the intensities of the cross-peaks in the 2D-NMR spectrum. Like in X-ray scattering an idea how the actual structure looks like is required.

 $\rightarrow$  constraint force field optimization of the atomic coordinates





### **NMR Structures (II)**

Due to the dynamic behavior in solution and limited time resolution of the NMR, an "averaged" structure is obtained.



Such .pdb files usually contain 10 individual solutions (models) and the resulting averaged structure.