Typical diseases

The search for pharmaceutical drugs used to be rather straight forward until recent times:

A wealth of information about the disease, its causes, and the clinical symptoms were readily available. Thus the starting point for the pharmacological therapy was known.

Example: inhibition of an enzyme

Thus the target was fixed. Frequently, experience with existing medications was available. Therefore a *valid target* or at least a *drugable target* was present.

→ The target undergoes a change of its activity caused by the drug

Often, multiple targets can be exploited to treat the same disease. → Therapeutic Categories



Therapeutic Categories

Grouping drugs under the aspect of their pharmacological and therapeutic application results in about 200 categories:

ACE Inhibitor
Adrenocortical Suppressant
Adrenocorticotropic Hormones
Aldose Reductase Inhibitors
Aldosterone Antagonists
α-adrenergic Agonists
α-adrenergic Blockers
α-Glucosidase Inhibitors
Anabolic Streroids

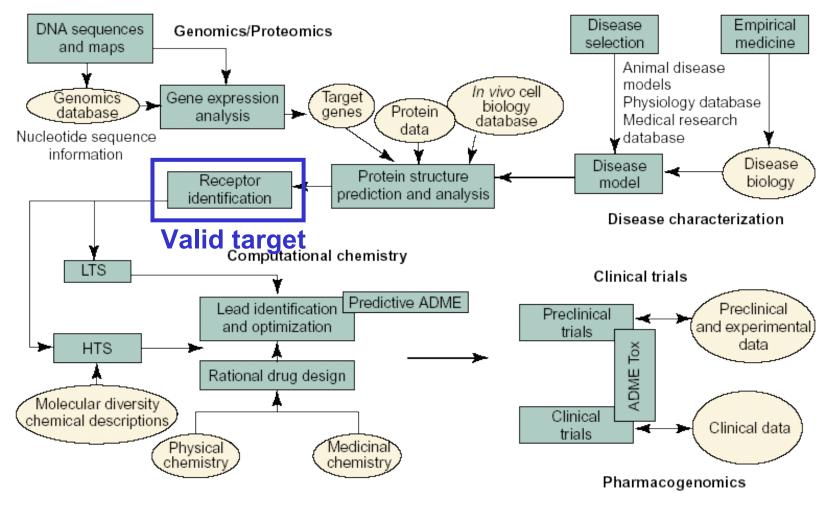
Analgesic, Dental
Analgesic, Narcotic
Analgesic, Non-narcotic
Androgens
Anesthetics, Inhaled
Anesthetics, Intravenous
Anesthetics, Local
Angiotensin II Antagonists
Anorexics

onists

see the corresponding section in the Merck Index

In most cases it is not obvious to recognize the treated disease from a therapeutic class (at least for non-medical persons), sometimes not even the actual molecular target(s).

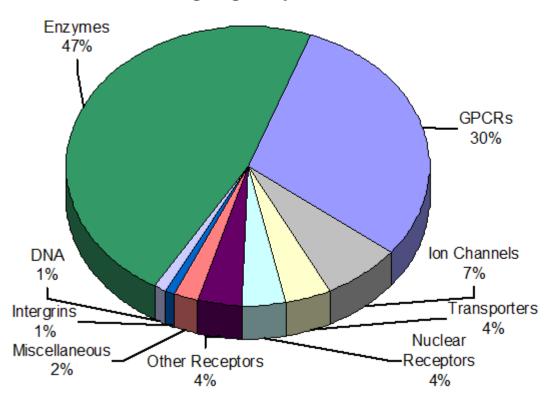
Flow of information in a drug discovery pipeline



Drug Discovery Today

typical targets

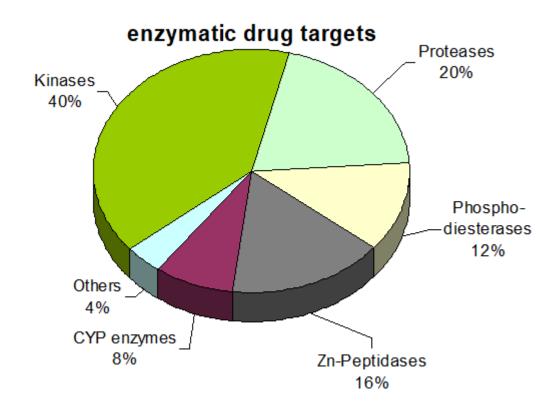
drug targets by biochemical class



Fractional content of marketed drugs according to their biochemical targets

data: Hopkins & Groom, Nat. Rev. Drug. Disc. 1 (2002) 727

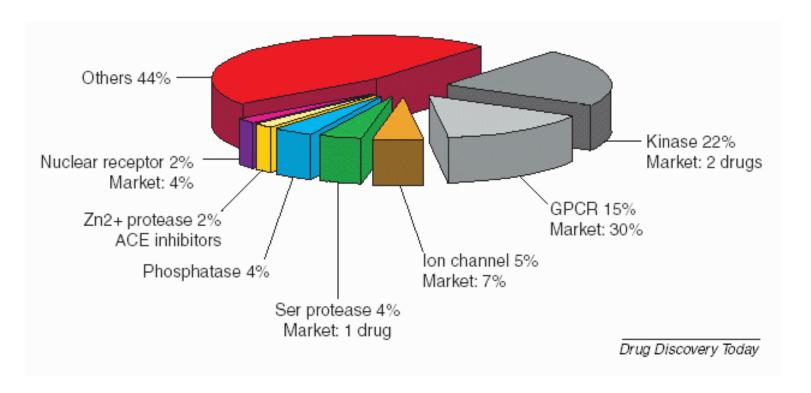
Enzymatic targets



Distribution within the class of enzymes

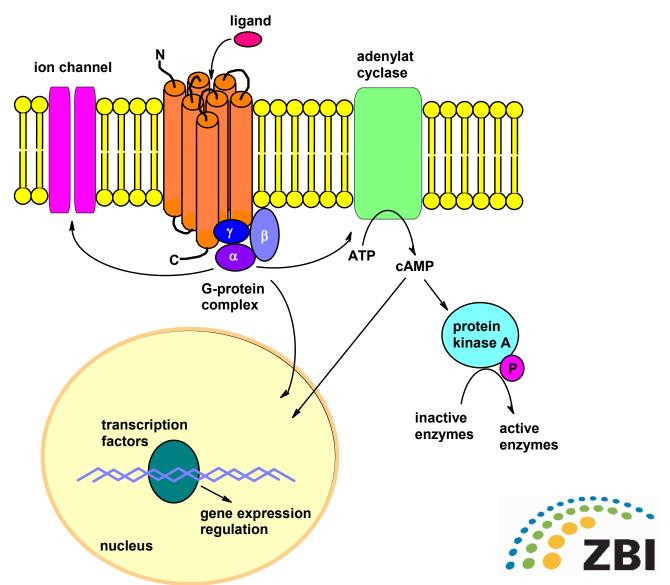


typical targets



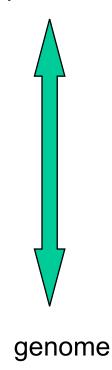
contribution to the human genome and marketed drugs about 500 enzymes have been used as targets 100,000 estimated potential targets in the genome

GPCRs and other targets



How do drugs interact with targets?

proteome



enzymes: substrate analogs, competitive ligands, reversible and irreversible inhibitors, allosteric modulators, protein-protein inhibitors

receptors: antagonists and agonists, orthosteric and allosteric ligands.

ion channels: openers and blockers (inhibitors)

transporters: inhibitors, e.g for (re-)uptake

nuclear receptors: binding to specific DNA-motives

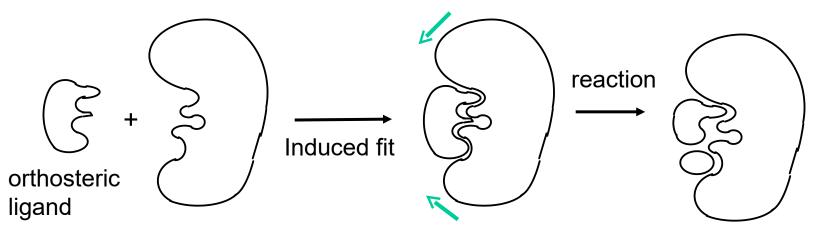
DNA: binding to groves, intercalation, alkylation, etc.

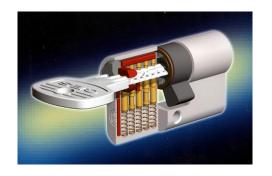
→ Possible targets can be found in a multitude of cell compartments and at different loci.

Problems: Drug delivery and drug transport / distribution

Drugs: mode of action (I)

Normal enzymatic turn-over





conformational change

lock and key principle



Drugs: mode of action (II)

competitive inhibitor:

higher affinity than natural substrate, directly acting at the orthosteric site



allosteric inhibitor/effector:

prevents binding by modifying the conformation, but at the allosteric site



Irreversible binding:

chemical reaction leads to inactivation of the enzyme

e.g. acetyl-salicylic acid acetylates Ser530 of Cyclooxygenase, ibrutinib reacts with Cys481 of Bruton's tyrosine kinase



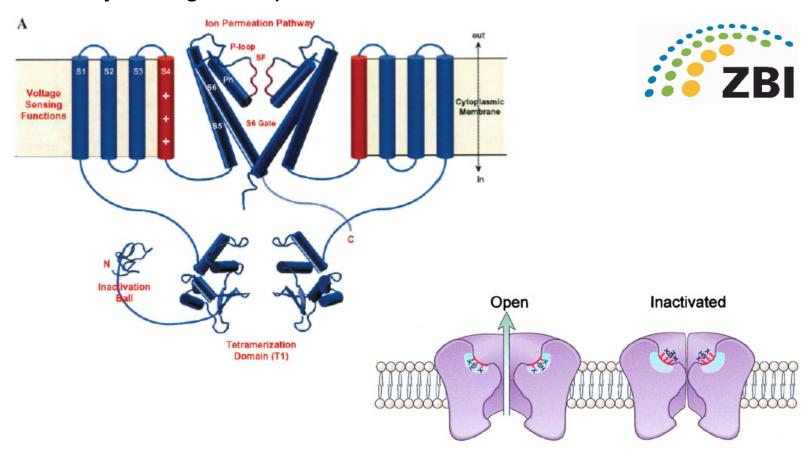
Competitive alternate ("wrong") substrate

e.g. methotrexate instead of dihydrofolate, antiviral nucleoside analoges

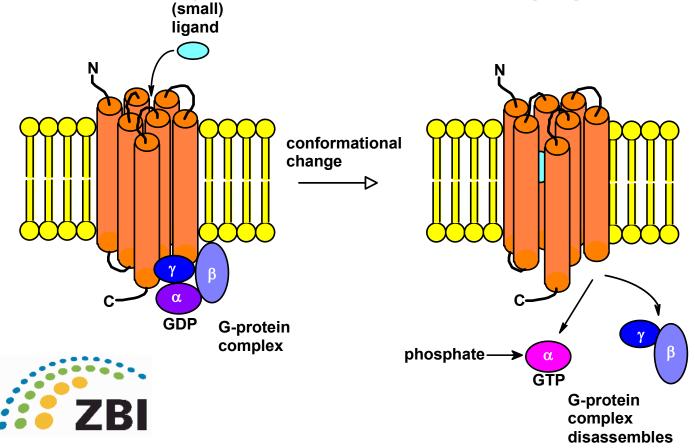


Drugs: mode of action (III)

Ion channels: Mode of action by ligand binding, and also indirectly through receptors



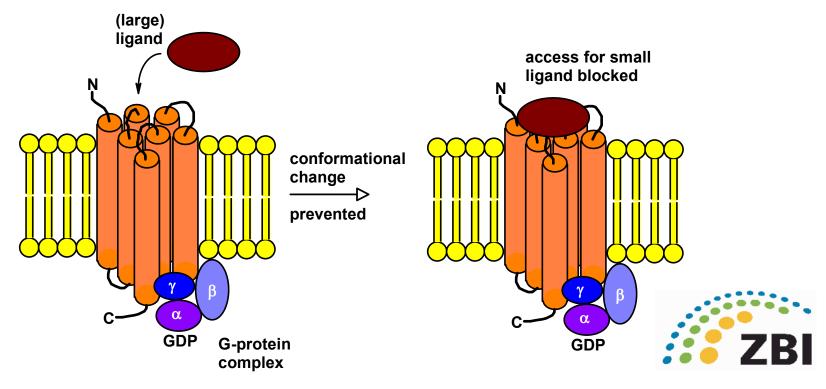
Drugs: mode of action (IV)



agonist: (orthosteric) ligand that causes an intrinsic effect (response of the receptor)

partial agonist: weakly working agonist with high binding affinity, thus also working as antagonist

Drugs: mode of action (V)

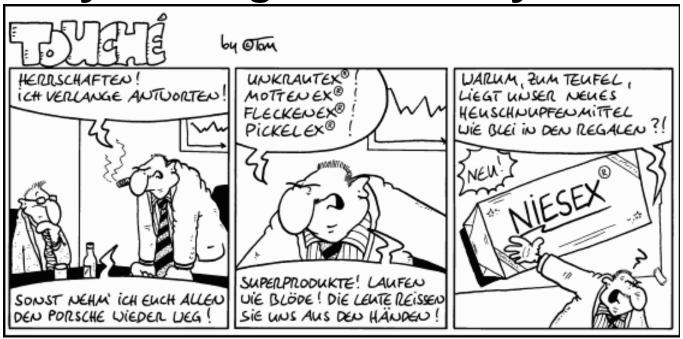


antagonist: ligand that prevents binding of the agonist, either directly (competitive binding at the orthosteric site) or indirectly (at allosteric site, prevents adoption of the reactive conformation)

inverse agonist: ligand stabilizing the inactive conformation

functional antagonist: prevents receptor response by a different mode of action

Why do drugs have funny names?



Examples for such faults in naming products exist!





Naming of drugs (I)

The **trade name** of a drug is usually chosen very carefully. Associative and speach-psychological aspects are considered.

Example within the german language:

The more x and y are appearing in the name, the more toxic.

Acetylsalicylsäure → Aspirin®

Problems will occur, if a product should get the same name throughout all countries. Examples:

Twix® (earlier: Raider)





Naming of drugs (II)

Furthermore, legal aspects have to be considered: existing words and words that imply a direct connection or target a specific consumer group cannot be protected.

Example: "Schülerschokolade" is not possible in Germany

Thus a lot of inspiration is required to find a pleasant sounding name. Frequently syllables and foreign words (latin, greek, spanish) are used that bear associations.





c.f. names for cars

® this name is approved and protected.

™ the producer indicates his intention to have this name protected.

Naming of drugs (III)

For the naming of the actual chemical substances there are also some (loose and empirical) guidelines.

Such names are adopted as "International Nonproprierary Name" (INN) or "United States Adopted Name" (USAN) at the lastest upon patent application.

Most of the time, the therapeutic class can be identified solely by the name. (similar names for substances with similar function.)

Prefixes and suffixes reflect chemical modification of the root compound.

Examples: ibufenac, clofenac, diclofenac, oxidanac

→ (unselective) Cyclooxygenase inhibitors

Naming of drugs (IV)

The World Health Organization (WHO) publishes updates regarding the use of stems in the selection of International Nonproprietary Names (INN) for pharmaceutical substances

Example: all drugs carrying the suffix (=stem) –coxib are selective Cycloxygenase II inhibitors:

celecoxib, cimicoxib, deracoxib, etoricoxib, firocoxib,

lumiracoxib, mavacoxib, parecoxib, robenacoxib, rofecoxib,

tilmacoxib, valdecoxib

In such cases the drug target is obvious.





Compound Databases

Size of typical substance libraries (2022)

ACD 12,000,000 chemicals

World Drug Index 80,000 compounds

USAN <10,000 in clinical trials

virtual library ≈100,000 compounds company, in house

PubChem > 112,000,000 compounds NCBI

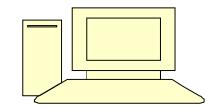
ChEMBL > 2,200,000 compounds EMBL

DrugBank > 500,000 drugs Uni. Alberta

ZINC15 >750,000,000 compounds UCSF

academic







Investment per new chemical entity: >800,000 \$
New chemical entities per year: ca. 15 strongly fluctuating

towards the drug (I)

disease model available symptoms medications Increasing knowledge applicable hypothesis therapeutic target of mechanism enzyme model cell model animal model transgenic animals sequenced genomes effort & expenses

Evolution of Disease Symptoms with Time

1800 1900 2000

Disease causing agents lack of hygiene germs, bacteria viruses



genetic disposition life style susceptibility





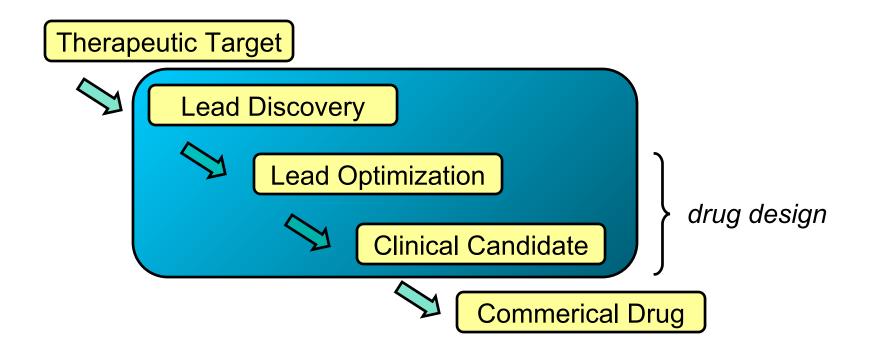
bioethic component

accepted legal definition of diseases

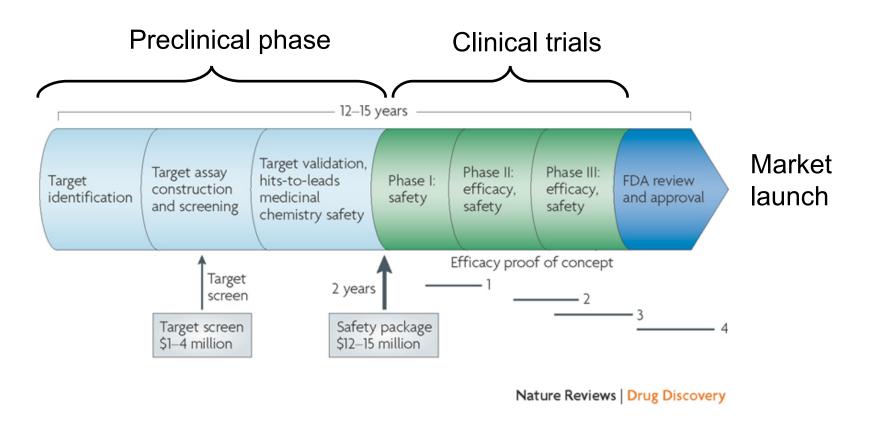


legal regulation for drug marketing (e.g. by the FDA)

The preclinical phase



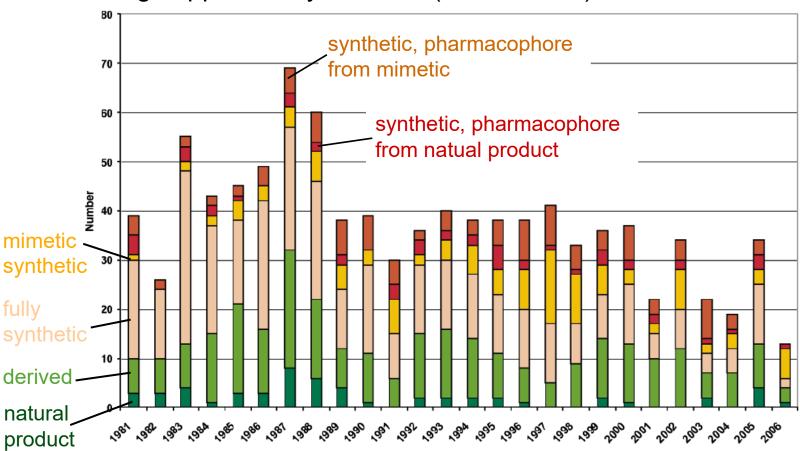
The drug discovery pipeline



A.D. Roses Nature Reviews Drug Discovery 7 (2008) 807.

Trend in approving new drugs (I)

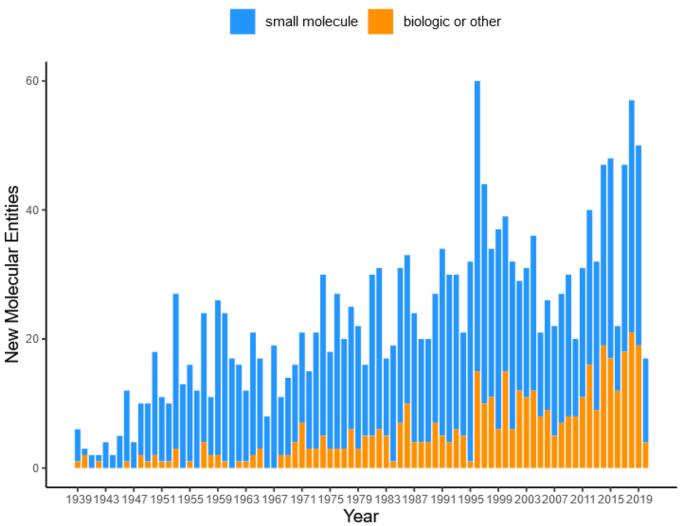
Drugs approved by the FDA (1981 - 2006)



Lit: D.J. Newman, G.M. Cragg *J.Nat.Prod.* **70** (2007) 461-477.

Trend in approving new drugs (II)





towards the drug (II)

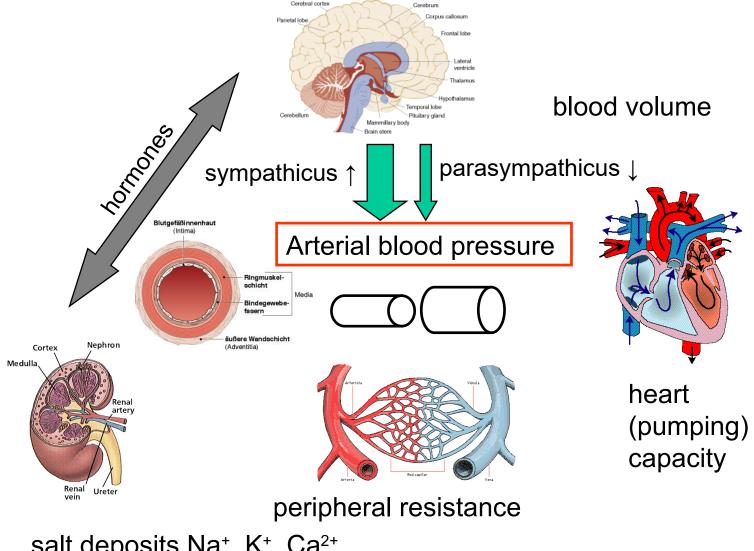
Example: arterial hypertension

Arterial hypertension [Arterielle Hypertonie] is a frequently observed condition (about 10 - 25% of all adults are affected). Persistent hypertension can lead to damage of blood vessels, the eyes, and the kidneys. → symptoms

category	systolic		diastolic	The state of the s
optimum	<120	and	<80	
normal	<130	and	<85	
normal-high	130 - 139	or	85 - 89	15/
mild HD	140 - 159	or	90 - 99	
moderate HD	160 - 179	or	100 - 109	
strong HD	>180	or	>110	mm (Hg)

source: Archives Int. Med. 157 (1997) 2413.

Regulation of the blood pressure (simplyfied)



salt deposits Na⁺, K⁺, Ca²⁺

diuretica and saluertica

lons in the blood and in other salt deposits bind water. By elimination of these ions the volume of the blood can be reduced.

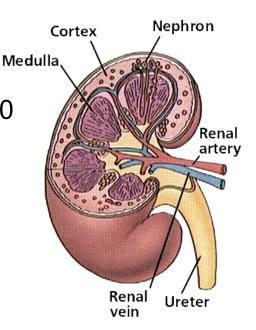
This effect is caused by diuretica and saluertica:

Examples: hydrochlorothiazide, furosemide

Therapeutic administration of thiazides since 1960

Disadvantages / side effects:

deficiency of potassium increased level of uric acid [Harnsäure] increased level of fatty acids in the serum not suitable with diabetes



α and β -blocker

Act relaxing via the peripheral nervous system and reduce the pumping capacity of the heart.

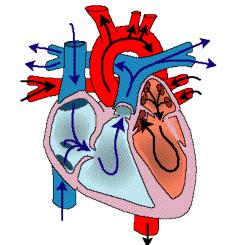
Examples: prasozin, tetrazosin, doxazosin, propanolol, atenolol, labetalol, pindolol

Simultaneously, the hormonal control is affected, whereby the peripheral resistance is diminished.

Therapeutic administration since 1970

Disadvantages and side effects:

withdrawl symptomes reduced capacity of the heart [Herzinsuffizienz] increased levels of fatty acids in the serum effects on the central nervous system



vasodilators and calcium antagonists

Act relaxing on the smooth muscles of the arterias and thereby reduce the resistance.

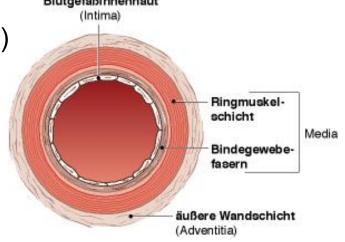
Bind to the *h*AT₂-receptor (vasodilators) or inhibit the calcium pump (calcium antagonists)

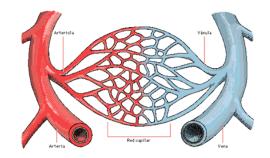
Examples: hydralazine, minoxidil, diazoxide, verapamil, diltiazem, nifedipine

Therapeutic administration since 1980

Disadvantages and side effects:

Predominately on the function of the heart





Angiotensin Coverting Enzyme Inhibitors

The endogenic oligopeptide Angiotensin II is one of the strongest vasoconstrictors.

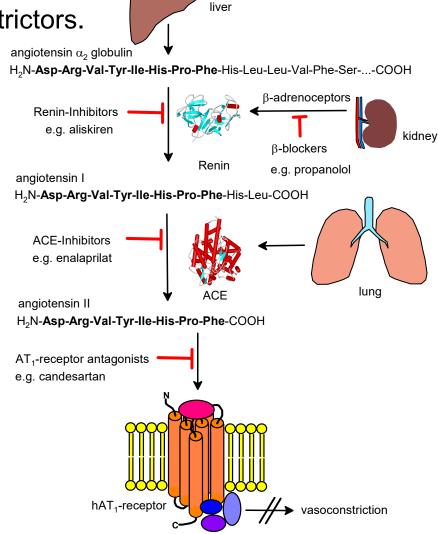
By inhibiting the angiotensin converting enzyme (ACE), the synthesis of Angiotensin II is disabled.

Examples: captopril, fosinopril, quinapril, enalapril

Therapeutic administration since 1990

disadvantages:

fetotoxic (pregnancy)



Angiotensin-II antagonists

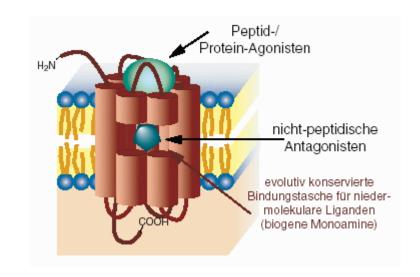
competitive binding of non-peptidic compounds to the *h*AT₁-receptor (GPCR), which is the binding site of Angiotensin II.

Examples: losartan, valsartan, irbesartan, candesartan, telmisartan, olmesartan, forsartan

therapeutic administration since 1995

disadvantages:

same as for ACE-inhibitors



Picture source: M. Gurrath Pharm. i. u. Zeit 288 (2001) 288.

"Evolution" of hypertension targets over time

targets therapeutic class

kidney diuretica, saluretica

nervous system α and β -blockers

calcium channels calcium channel blockers

(adrenal gland)

*h*AT₂-receptor vasodilators

ACE ACE-inhibitors

*h*AT₁-receptor Angiotensin II antagonists

increasing specificity

Methods for Determining Atomic Structures

X-ray and electron microscopy

Diffraction and scattering of electromagnetic waves, respectively electrons

Pro: high resolution Con: "artificial" conditions

solid state, crystal

NMR (nuclear magnetic resonance)

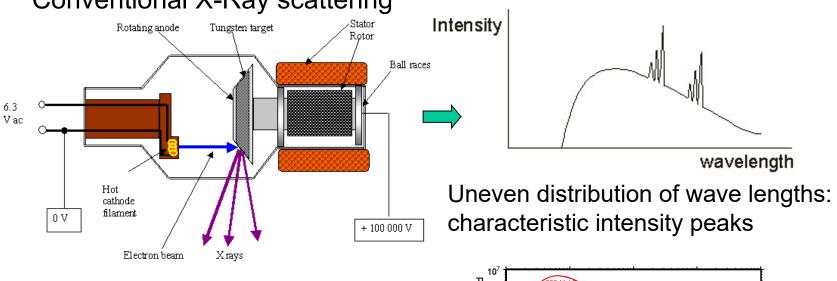
Absorption of electromagnetic waves

Pro: "natural" conditions aqueous solution

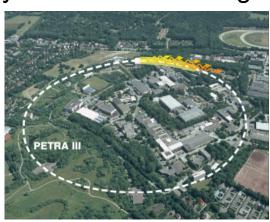
Con: only for small proteins <100 amino acids

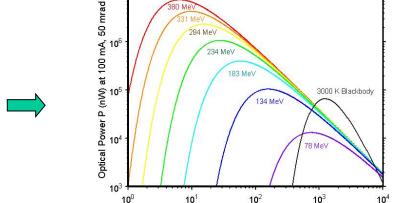
Obtaining X-Ray structures (I)

Conventional X-Ray scattering



Synchrotron scattering



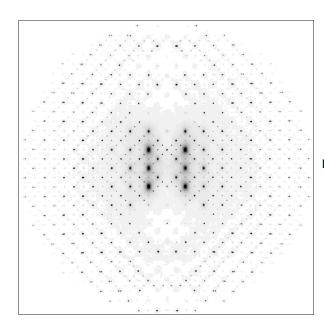


Continuous distribution of wave lengths

Wavelength λ (nm)

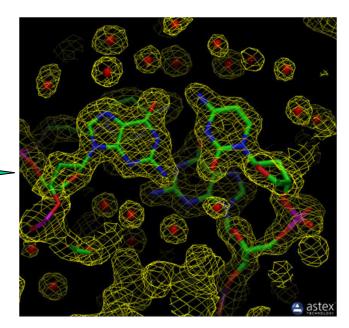
Obtaining X-Ray structures (II)

The arrangement of atoms in the crystal gives rise to a diffraction pattern



Mathematical transformation &

Idea how the actual structure looks like



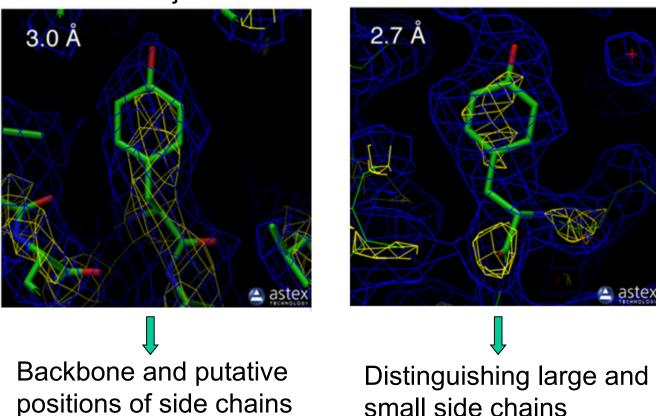
Electron density map

3D coordinates can be obtained from www.rcsb.org (Protein Data Bank) and electron density maps from http://www.ebi.ac.uk/pdbe/

Accurracy of X-Ray structures (I)

The resolution given in .pdb files tells us the smallest wavelength (in Å) the crystal was able to diffract.

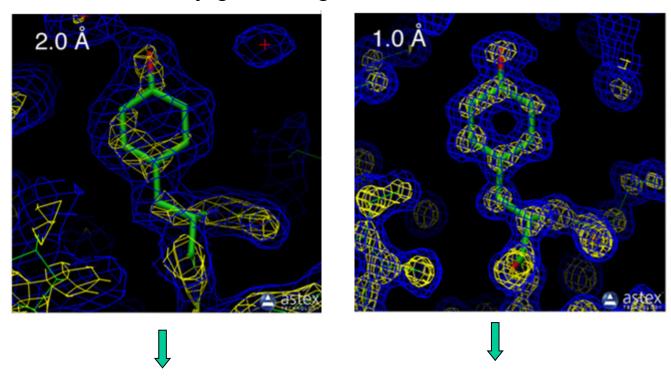
→ We "see" objects of that size



small side chains

Accurracy of X-Ray structures (II)

The atomic coordinates are fitted into the electron density grid using a force field



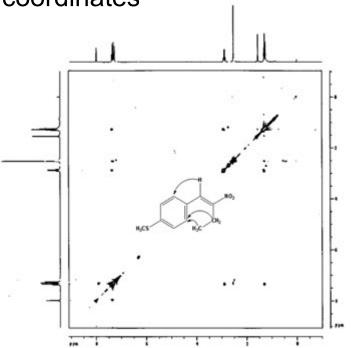
Different rotamers of the side chains can be assigned

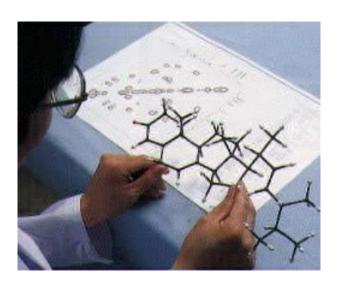
Atomic coordinates can be assigned unambigously

NMR Structures (I)

The distance between atoms can be derived from the intensities of the cross-peaks in the 2D-NMR spectrum. Like in X-ray scattering an idea how the actual structure looks like is required.

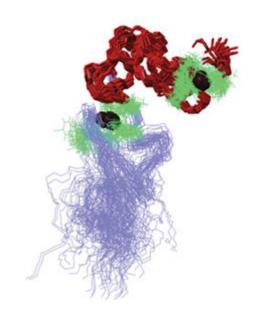
→ constraint force field optimization of the atomic coordinates





NMR Structures (II)

Due to the dynamic behavior in solution and limited time resolution of the NMR, an "averaged" structure is obtained.



Such .pdb files usually contain 10 individual solutions (models) and the resulting averaged structure.