Therapeutic Categories

Grouping drugs under the aspect of their pharmacological and therapeutic application results in about 200 categories:

- ACE Inhibitor Adrenocortical Suppressant Adrenocorticotropic Hormones Aldose Reductase Inhibitors Aldosterone Antagonists α -adrenergic Agonists α -adrenergic Blockers α -Glucosidase Inhibitors Anabolic Streroids
- ...c.f. the Merck Index

Analgesic, Dental Analgesic, Narcotic Analgesic, Non-narcotic Androgens Anesthetics, Inhaled Anesthetics, Intravenous Anesthetics, Local Angiotensin II Antagonists Anorexics





In most cases it is not obvious to conclude the treated disease from a therapeutic class. (At least for non-medical persons)

Typical diseases

The search for pharmaceutical drugs used to be rather straight forward until recent times:

A wealth of information about the disease, its causes, and the clinical symptoms were readily available. Thus the starting point for the pharmacological therapy was known.

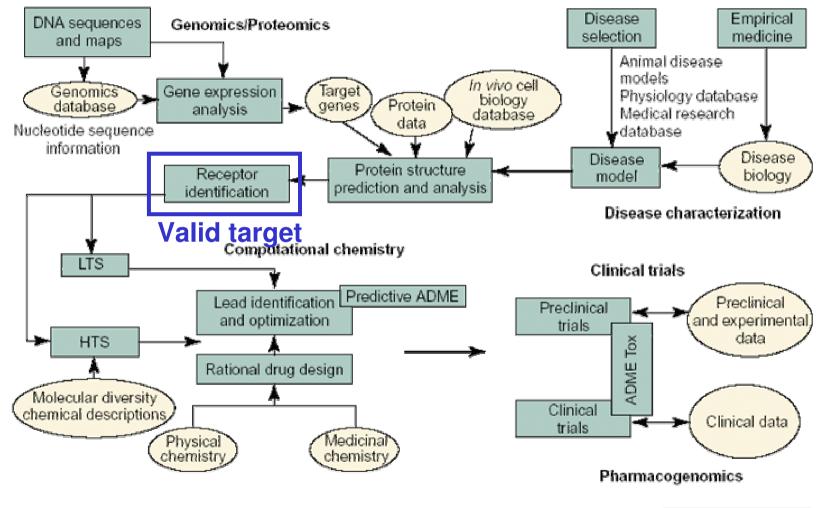
Example: inhibition of an enzyme

Thus the target was fixed. Frequently, experience with existing medications was available. Therefore a *valid target* or at least a *drugable target* was present.

 \rightarrow The *target* undergoes a change of its activity caused by the drug



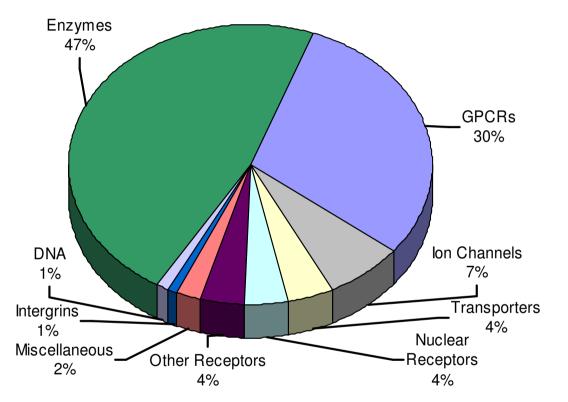
Flow of information in a drug discovery pipeline



Drug Discovery Today

typical targets

drug targets by biochemical class

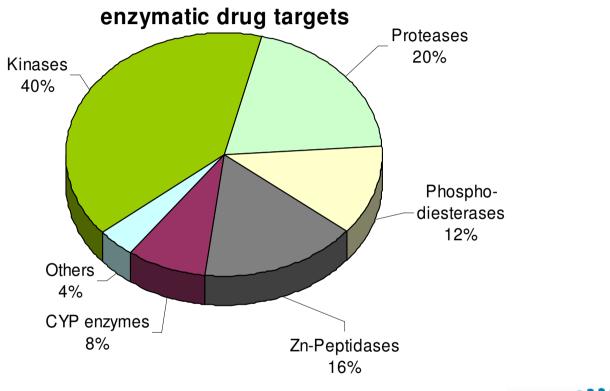


Fractional content of marketed drugs according to their biochemical targets

data: Hopkins & Groom, Nat. Rev. Drug. Disc. 1 (2002) 727



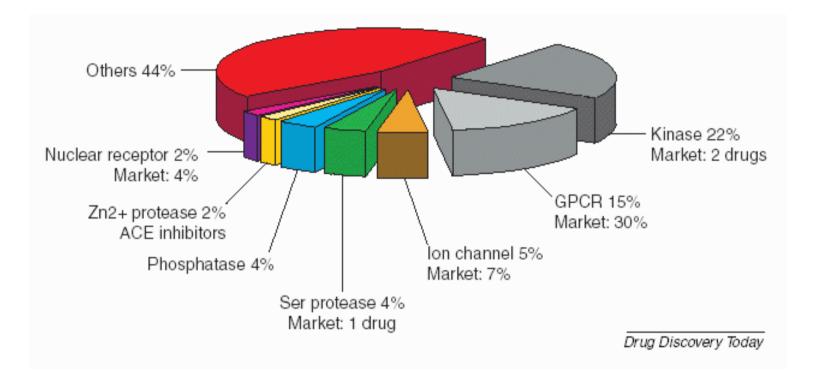
Enzymatic targets



Distribution within the class of enzymes



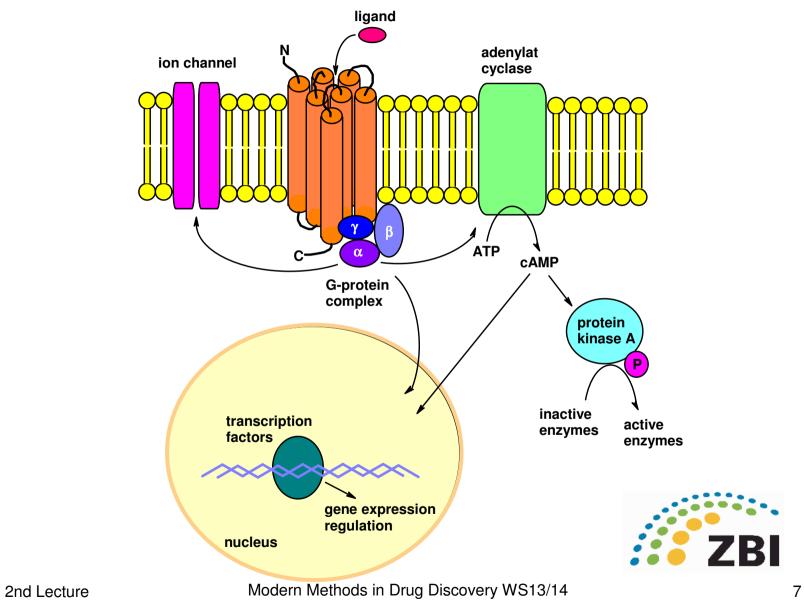
typical targets



contribution to the human genome and marketed drugs about 500 enzymes have been used as targets 100,000 estimated potential targets in the genome



GPCRs and other targets



How do drugs interact with targets ?

proteome

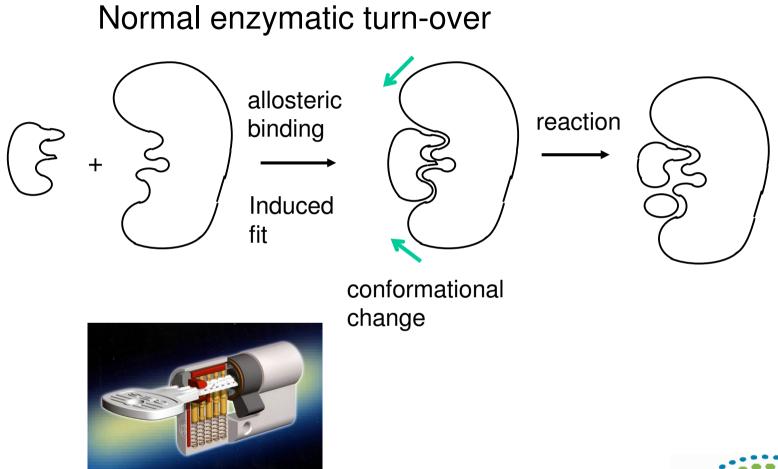
enzymes: substrate analogs, competitive ligands, reversible and irreversible inhibitors
receptors: antagonists and agonists
ion channels: openers and blockers (inhibitors)
transporters: (re-)uptake inhibitors
DNA / nuclear receptors: intercalate, binding to specific DNA-motives, groves, etc.

genome

 \rightarrow Possible targets can be found in a multitude of cell compartments and at different loci.

Problems: Drug delivery and drug transport / distribution

Drugs: mode of action (I)





Drugs: mode of action (II)

competitive inhibitor:

higher affinity than natural substrate, directly acting



allosteric inhibitor/effector:

prevents binding by modifying the conformation



Irreversible binding:

chemical reaction leads to inactivation of the enzyme

e.g. acetyl-salicylic acid acetylates Ser530 of Cyclooxygenase



Anti-metabolite:

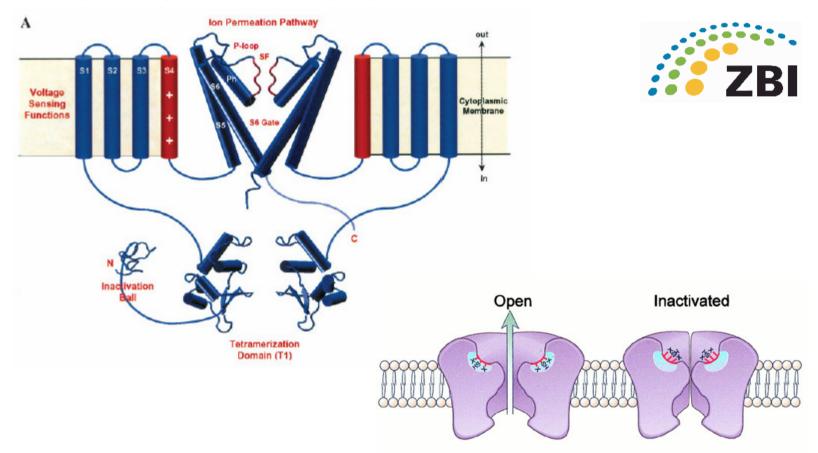
Competitive alternate ("wrong") substrate

e.g. methotrexate instead of dihydrofolate, antiviral nucleoside analoges

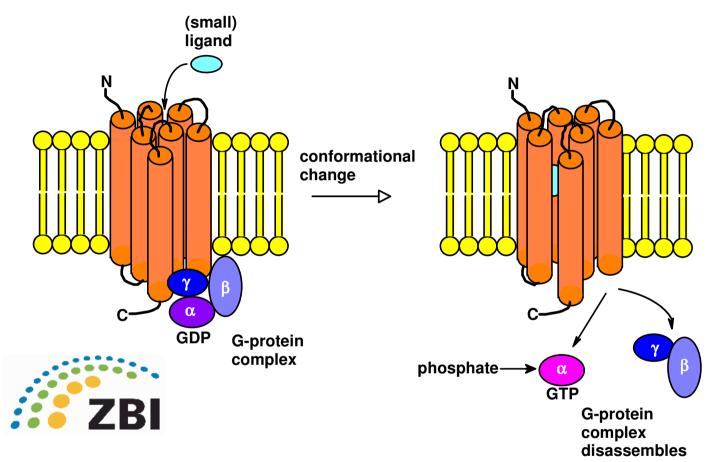


Drugs: mode of action (III)

Ion channels: Mode of action by ligand binding, indirectly through receptors, or voltage gated



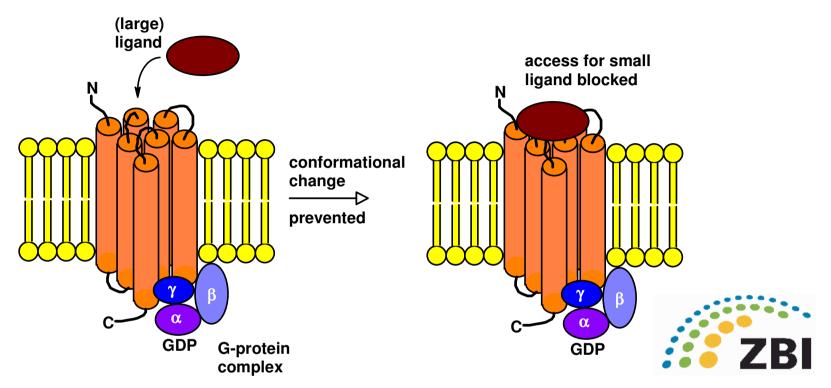
Drugs: mode of action (IV)



agonist: ligand that causes an intrinsic effect (response of the receptor)

partial agonist: weakly working agonist with high binding affinity, thus also working as antagonist

Drugs: mode of action (V)



antagonist: ligand that prevents binding of the agonist, either directly (competitive binding) or indirectly (allosteric, prevents adoption of the reactive conformation)

inverse agonist: ligand stabilizing the inactive conformation

functional antagonist: prevents receptor response by a different mode of action Modern Methods in Drug Discovery WS13/14 2nd Lecture

Why do drugs have funny names ?



Examples for such faults in naming products exist !





Naming of drugs (I)

The trade name of a drug is usually chosen very carefully. Associative and speach-psychological aspects are considered.

Example within the german language:

The more x and y are appearing in the name, the more toxic.

 $Acetylsalicylsäure \rightarrow Aspirin \textcircled{B}$

Problems will occur, if a product should get the same name throughout all countries. Examples:

Twix® (earlier: Raider)





Naming of drugs (II)

Furthermore, legal aspects have to be considered: existing words and words that imply a direct connection or target a specific consumer group cannot be protected.

Example: "Schülerschokolade" is not possible in Germany

Thus a lot of inspiration is required to find a pleasant sounding name. Frequently syllables and foreign words (latin, greek, spanish) are used that bear associations.





c.f. names for cars

® this name is approved and protected.

[™] the producer indicates his intention to have this name protected.

Naming of drugs (III)

For the naming of the actual chemical substances there are also some (loose and empirical) guidelines.

Such names are adopted as "International Nonproprierary Name" (INN) or "United States Adopted Name" (USAN) at the lastest upon patent application.

Most of the time, the therapeutic class can be identified solely by the name. (similar names for substances with similar function.)

Prefixes and suffixes reflect chemical modification of the root compound.

Examples: ibufenac, clofenac, diclofenac, oxidanac

Naming of drugs (IV)

The World Health Organization (WHO) publishes updates regarding the use of stems in the selection of International Nonproprietary Names (INN) for pharmaceutical substances

Example: all drugs carrying the suffix (=stem) –coxib are selective cycloxygenase inhibitors:

celecoxib, cimicoxib, deracoxib, etoricoxib, firocoxib, lumiracoxib, mavacoxib, parecoxib, robenacoxib, rofecoxib, tilmacoxib, valdecoxib

In such cases the drug target is obvious.



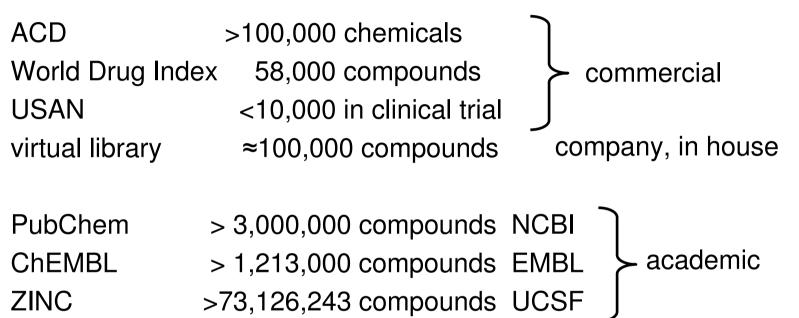


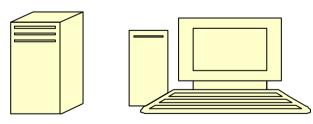
http://www.2l.no/2L23details.htm Modern Methods in Drug Discovery WS13/14

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Compound Databases

existing substance libraries

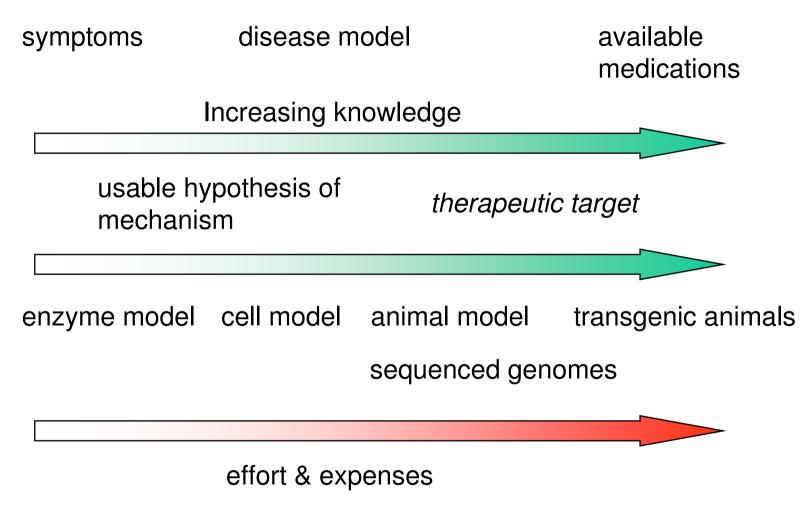




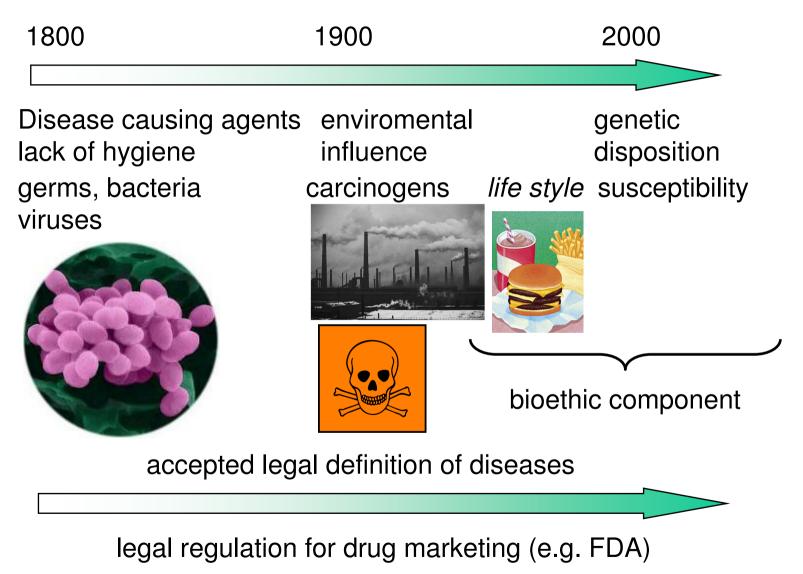


Investment per new chemical entity: >500,000 \$ New chemical entities per year: ca. 15 Modern Methods in Drug Discovery WS13/14

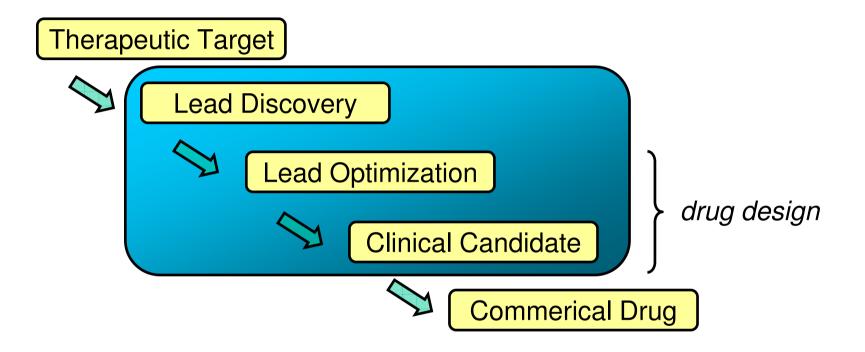
towards the drug (I)



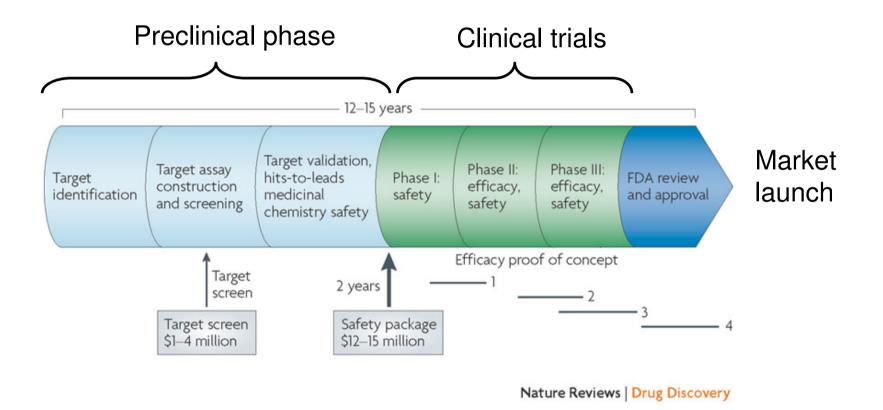
Evolution of Disease Symptoms with Time



The preclinical phase



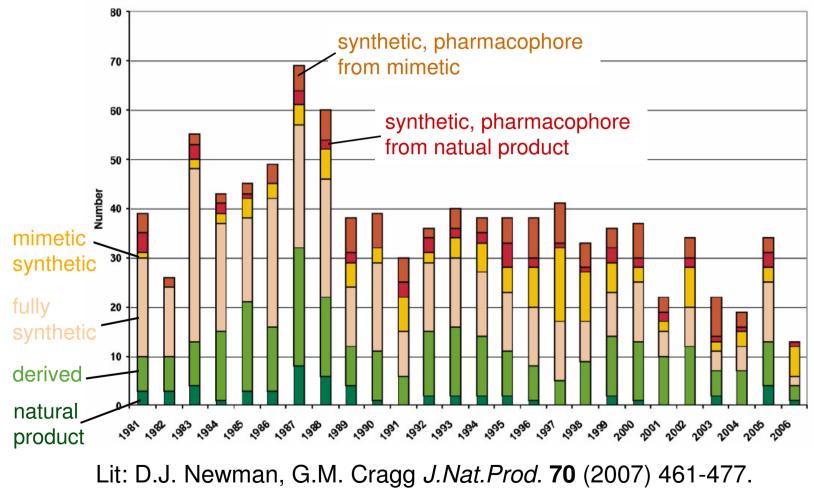
The drug discovery pipeline



A.D. Roses Nature Reviews Drug Discovery 7 (2008) 807.

Trend in approving new drugs

Drugs approved by the FDA within the last 25 years



towards the drug (II)

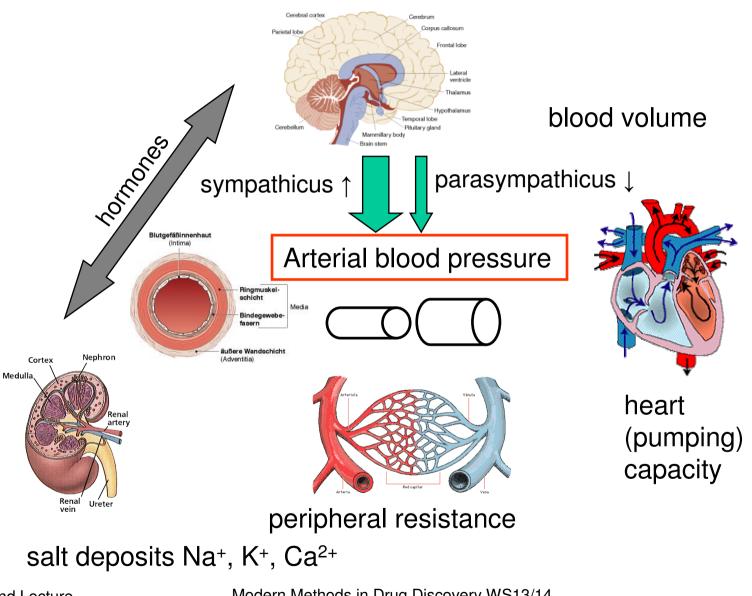
Example: arterial hypertension

Arterial hypertension [Arterielle Hypertonie] is a frequently observed condition (about 10 - 25% of all adults are affected). Persisent hypertension can lead to damage of blood vessels, the eyes, and the kidneys. \rightarrow symptoms

category	systolic		diastolic	
optimum	<120	and	<80	
normal	<130	and	<85	
normal-high	130 - 139	or	85 - 89	
mild HD	140 - 159	or	90 - 99	
moderate HD	160 - 179	or	100 - 109	
strong HD	>180	or	>110	mm (Hg)



source: Archives Int. Med. 157 (1997) 2413.



Regulation of the blood pressure (simplyfied)

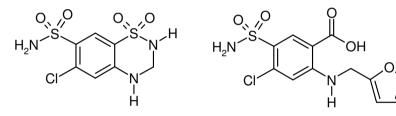
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diuretica and saluertica

lons in the blood and in other salt deposits bind water. By elimination of these ions the volume of the blood can be reduced.

This effect is caused by diuretica and saluertica:

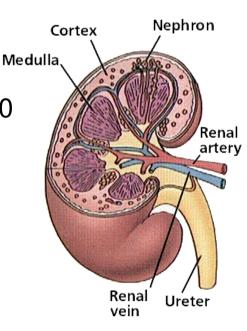
Examples: hydrochlorothiazide, furosemide



Therapeutic administration of thiazides since 1960

Disadvantages / side effects:

deficiency of potassium increased level of uric acid [Harnsäure] increased level of fatty acids in the serum not suitable with diabetes



α and $\beta\text{-blocker}$

Act relaxing via the peripheral nervous system and reduce the pumping capacity of the heart.

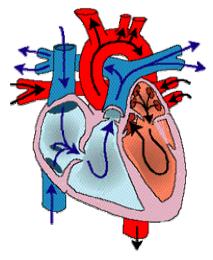
Examples: prasozin, tetrazosin, doxazosin, propanolol, atenolol, labetalol, pindolol

Simultaneously, the hormonal control is affected, whereby the peripheral resistance is diminished.

Therapeutic administration since 1970

Disadvantages and side effects:

withdrawl symptomes reduced capacity of the heart [Herzinsuffizienz] increased levels of fatty acids in the serum effects on the central nervous system



vasodilators and calcium antagonists

Act relaxing on the smooth muscles of the arterias and thereby reduce the resistance.

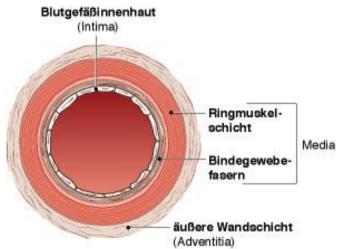
Bind to the hAT_2 -receptor or inhibit the calcium pump

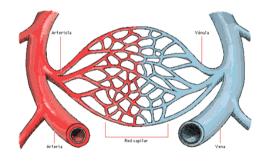
Examples: hydralazine, minoxidil, diazoxide, verapamil, diltiazem, nifedipine

Therapeutic administration since 1980

Disadvantages and side effects:

Predominately on the function of the heart





Angiotensin Coverting Enzyme Inhibitors

The endogenic oligopeptide Angiotensin II is one of the strongest vasoconstrictors. By inhibiting the angiotenisn converting enzyme (ACE) the synthesis of Angiotensin II is disabled.

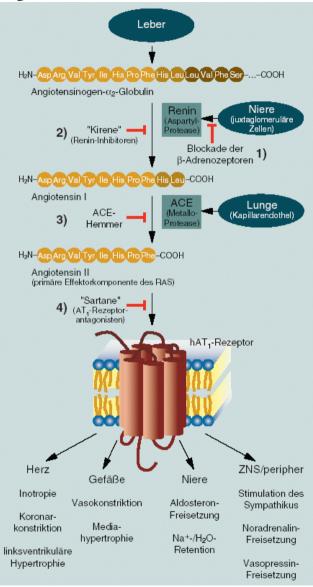
Examples: captopril, fosinopril, quinapril

Therapeutic administration since 1990

disadvantages:

fetotoxic (pregnancy)

Picture source: M. Gurrath *Pharm. i. u. Zeit* **288** (2001) 288.



Angiotensin-II antagonists

competitive binding of non-peptidic compounds to the hAT_1 -receptor (GPCR), which is the binding site of Angiotensin II.

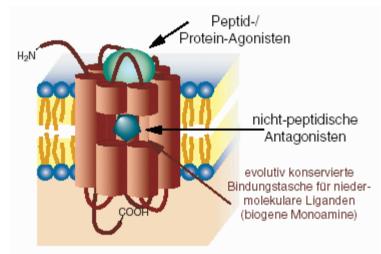
Examples: losartan, valsartan, irbesartan, candesartan, telmisartan

Furthermore in clinical testing: olmesartan, forsartan

therapeutic administration since 1995

disadvantages:

same as for ACE-inhibitors



Picture source: M. Gurrath Pharm. i. u. Zeit 288 (2001) 288.

Evolution of targets over time

targets	therapeutic class		
kidney	diuretica, saluretica		ing ity
nervous system	α and β -blocker		eas cific
hAT ₂ -receptor	vasodilators		incı spe
ACE	ACE-inhibitors	$\langle \rangle$	
hAT ₁ -receptor	Angiotensin II antagonists	V	

Methods for Determining Atomic Structures

X-ray and electron microscopy

Diffraction and Scattering of electromagnetic waves, respectively electrons

Pro: resolution

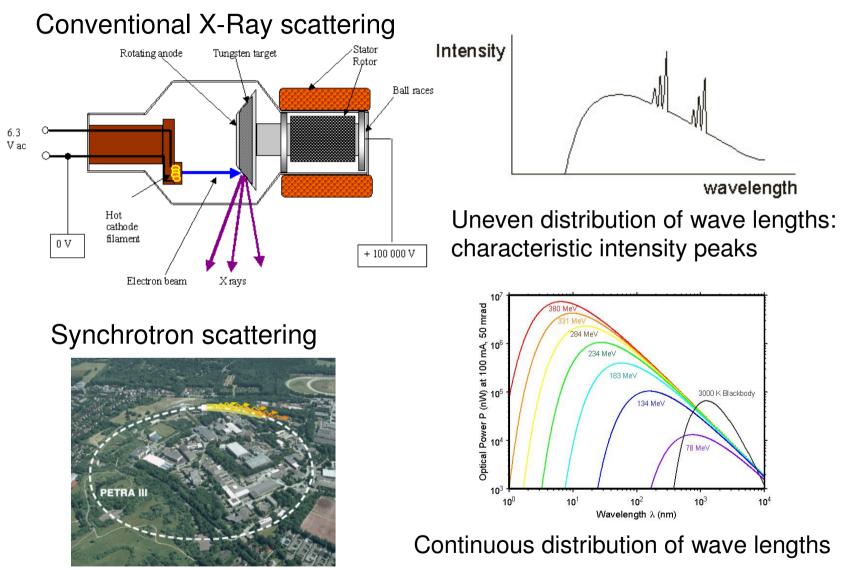
Con: "artificial" conditions

NMR (nuclear magnetic resonance)

Absorption of electromagnetic waves

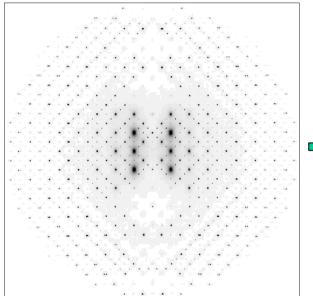
Pro: "natural" conditions Con: only for small proteins

Obtaining X-Ray structures (I)



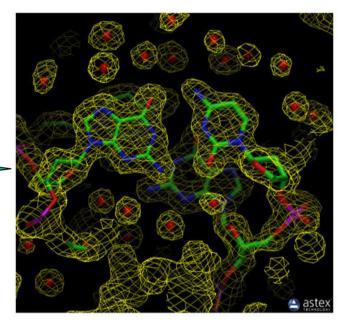
Obtaining X-Ray structures (II)

The arrangement of atoms in the crystal gives rise to a diffraction pattern



Mathematical transformation &

Idea how the actual structure looks like

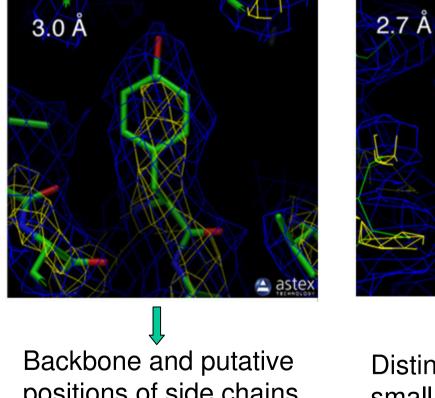


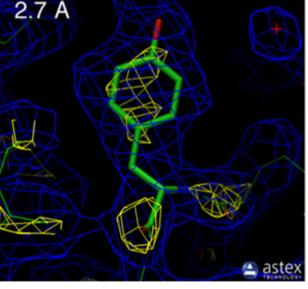
Electron density map

Accurracy of X-Ray structures (I)

The resolution given in .pdb files tells us the smallest wavelength (in Å) the crystal was able to diffract.

 \rightarrow We "see" objects of that size



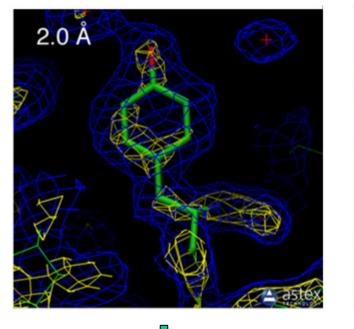


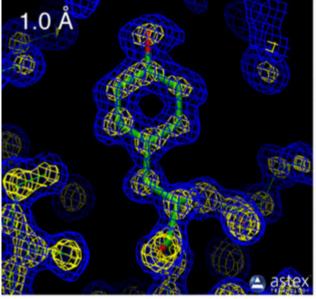
positions of side chains

Distinguishing large and small side chains

Accurracy of X-Ray structures (II)

The atomic coordinates are fitted into the electron density grid using a force field





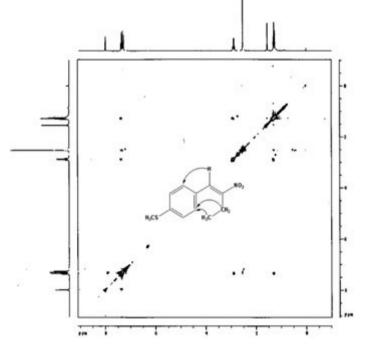
Different rotamers of the side chains can be assigned

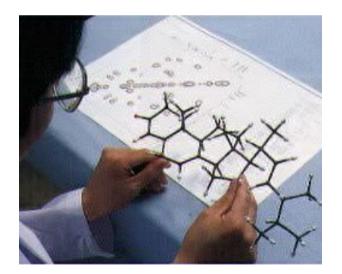
Atomic coordinates can be assigned unambigously

NMR Structures (I)

The distance between atoms can be derived from the intensities of the cross-peaks in the 2D-NMR spectrum. Like in X-ray scattering an idea how the actual structure looks like is required.

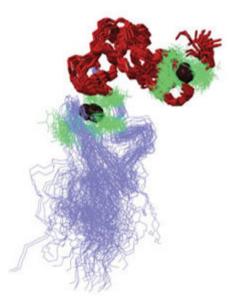
 \rightarrow constraint force field optimization of the atomic coordinates





NMR Structures (II)

Due to the dynamic behavior in solution and limited time resolution of the NMR, an "averaged" structure is obtained.



Such .pdb files usually contain 10 individual solutions and the resulting averaged structure.

2nd Lecture